

PSD '20 -- Xray Lecture 5, 6

Patterson Space, Molecular Replacement and Heavy Atom Isomorphous Replacement

$$\rho(r) = \sum_h |F(h)| e^{i\alpha(h)} e^{-i2\pi h \cdot r}$$

The Phase Problem

We can't measure the phases!

X-ray detectors (film, photomultiplier tubes, CCDs, etc) **can measure only the intensity of the X-rays** (which is the *amplitude squared*), but we need the full wave equations $Ae^{i\alpha}$ for each reflection to do the reverse Fourier transform.

And because it is called the phase "*problem*", the process of getting the phases is called a "solution". That's why we say we "solved" the crystal structure, instead of "*measured*" or "*determined*" it.

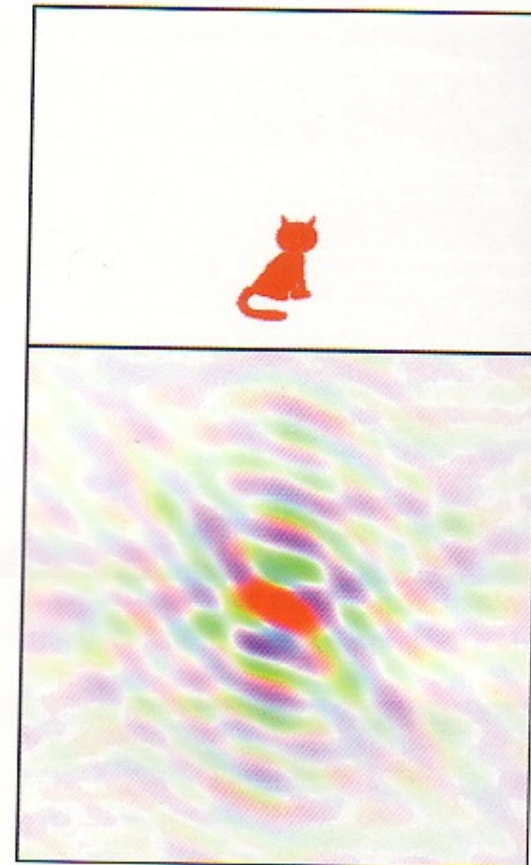
Phase is more important than amplitude

color=phase angle
darkness=amplitude

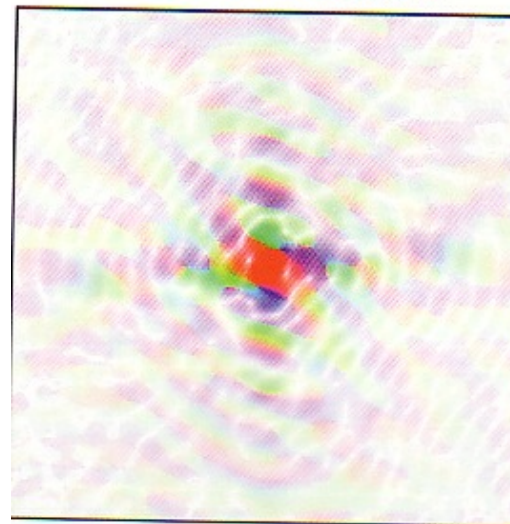
Combining the amplitudes of the duck with the phases of the cat, then reversing the FT, we get...the cat.



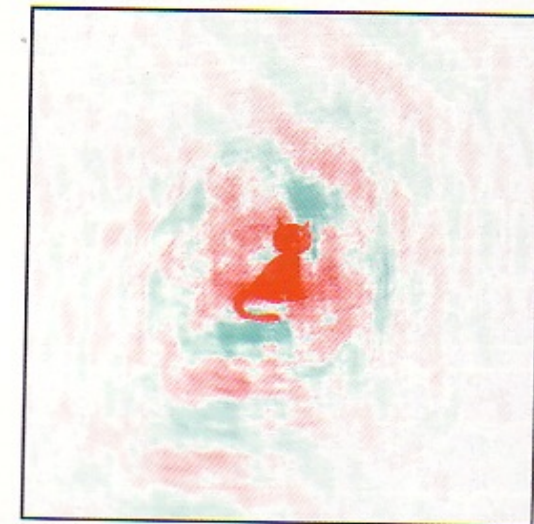
a Duck and duck FT



b Cat and cat FT



c Duck intensities and cat phases

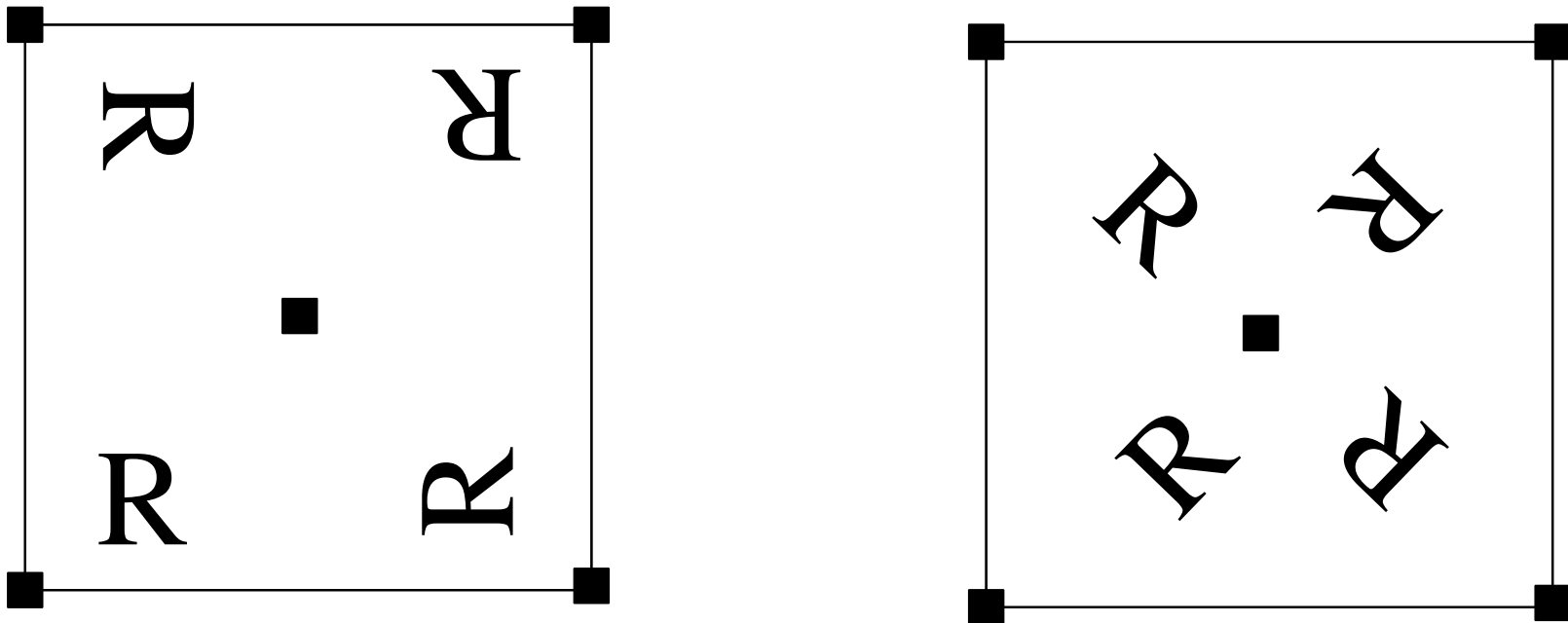


d Back-transform of *c*

molecular replacement

If the structure of the molecule is known approximately, then the phases can be calculated.

BUT. We need to know how the molecule is oriented.



Same molecule. Same symmetry. Same cell dimensions.
But... these two crystals are not isomorphous.

We can use *homology* to infer structure

If the protein sequence identity is > 25%, then infer that the sequences are "homologous".

Homologous proteins have similar structures.

How similar is not known until both structures are solved.

Molecular replacement will not work if the structures are too different.

If a homolog of known structure exists, then it can be used to do molecular replacement

The unknown is the *orientation*

The space of all possible *rigidbody* transformations of a molecule has 6 dimensions. 3 angles of rotation (defining a matrix of 9 coefficients), and a 3D translation vector.

$$x' = c_{11}x + c_{21}y + c_{31}z + v_x$$

$$y' = c_{12}x + c_{22}y + c_{32}z + v_y$$

$$z' = c_{13}x + c_{23}y + c_{33}z + v_z$$

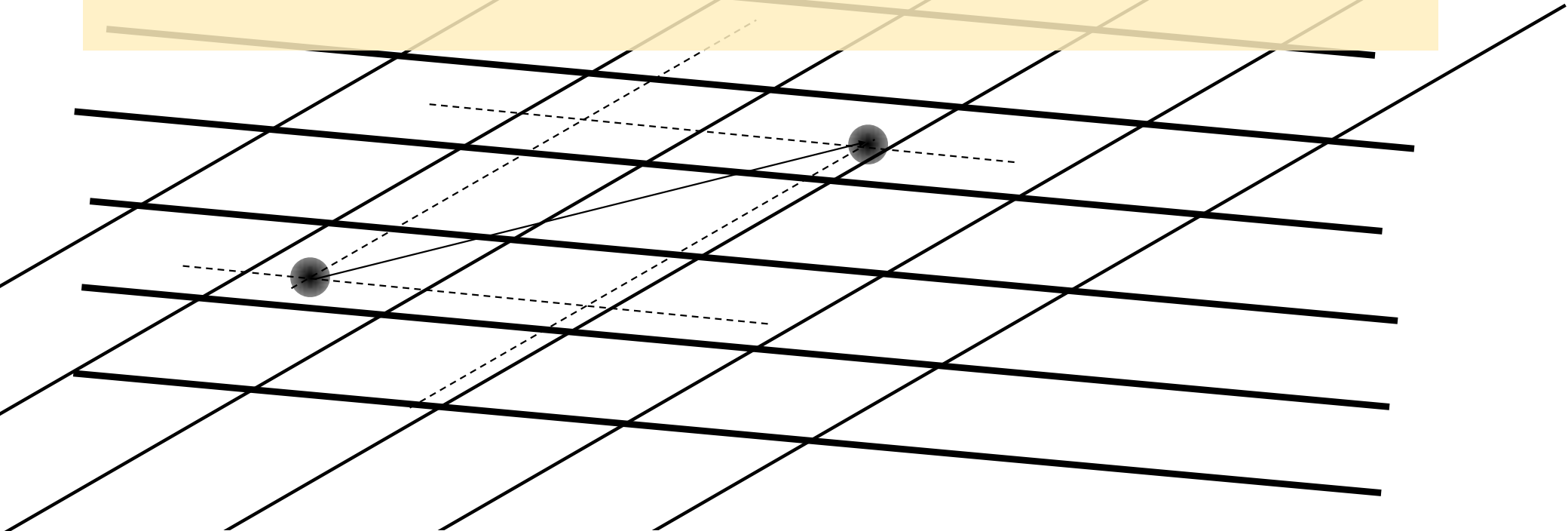
$$\text{or } \mathbf{x}' = \underline{\mathbf{C}}\mathbf{x} + \mathbf{v}$$

Therefore, the position of our molecule in the crystal unit cell must be a 6D transformation of its current position. Molecular replacement is the method for finding the angles and vector that define the transformation.

Patterson Space

Consider the following...

Amplitudes for a 2 atom crystal are maximal when the atoms align with the Bragg planes



This of two symmetry-related heavy atoms....

The amplitude of a wave $F(h\ k\ l)$ scattered by just two atoms depends how similar are their phases. If they have the same phase, any phase it doesn't matter, then the amplitude is greatest.

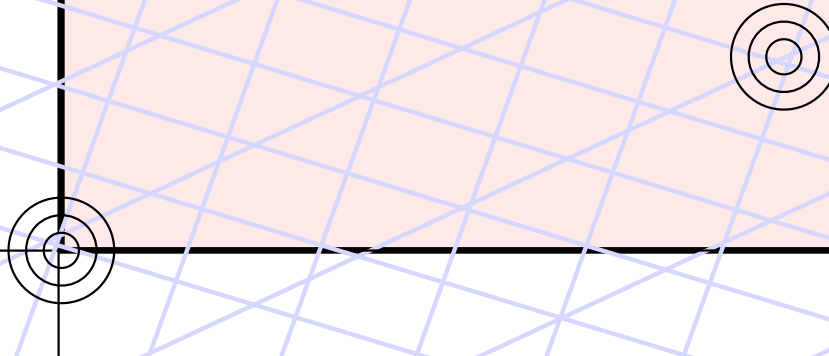


Constructing a Patterson map

If these planes have the largest amplitudes, the atoms must be
— at the intersection of these Bragg planes.

Does not require phases!!

Phase zero Fourier transform



If we calculate the inverse FT with all phases set to zero, the “wave crests” of electron density all go through the origin.

But the other places where wave crests intersect, are the relative positions of heavy atoms.

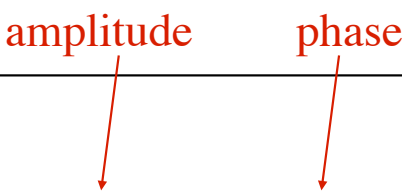
The phase zero inverse Fourier is called a **Patterson Map**

Reverse Fourier Transform vs Patterson Function


Reverse FT *with*
phases

$$\rho(r) = \sum_h |F(h)| e^{i\alpha(h)} e^{-i2\pi h \cdot r}$$

amplitude phase



Reverse FT *without*
phases

$$\rho(r) = \sum_h |F(h)| e^{-i2\pi h \cdot r}$$


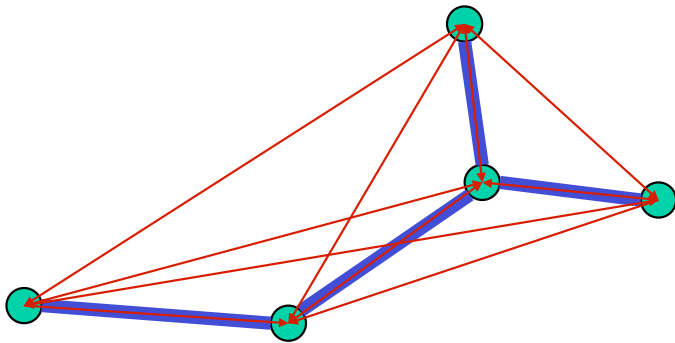
This is the *Patterson function*

it uses the measured
amplitude, **no phase**

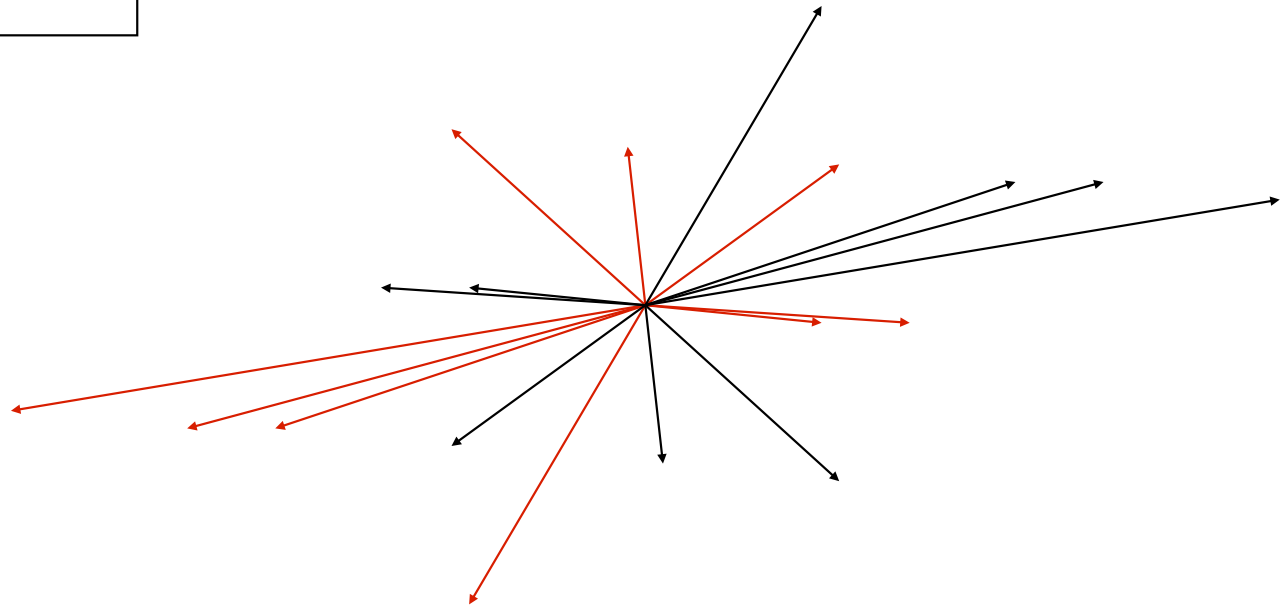
Patterson map represents all inter-atomic vectors

To generate a centro-symmetric projection in 2D, draw all inter-atomic vectors, then move the tails to the origin. The heads are where peaks would be.

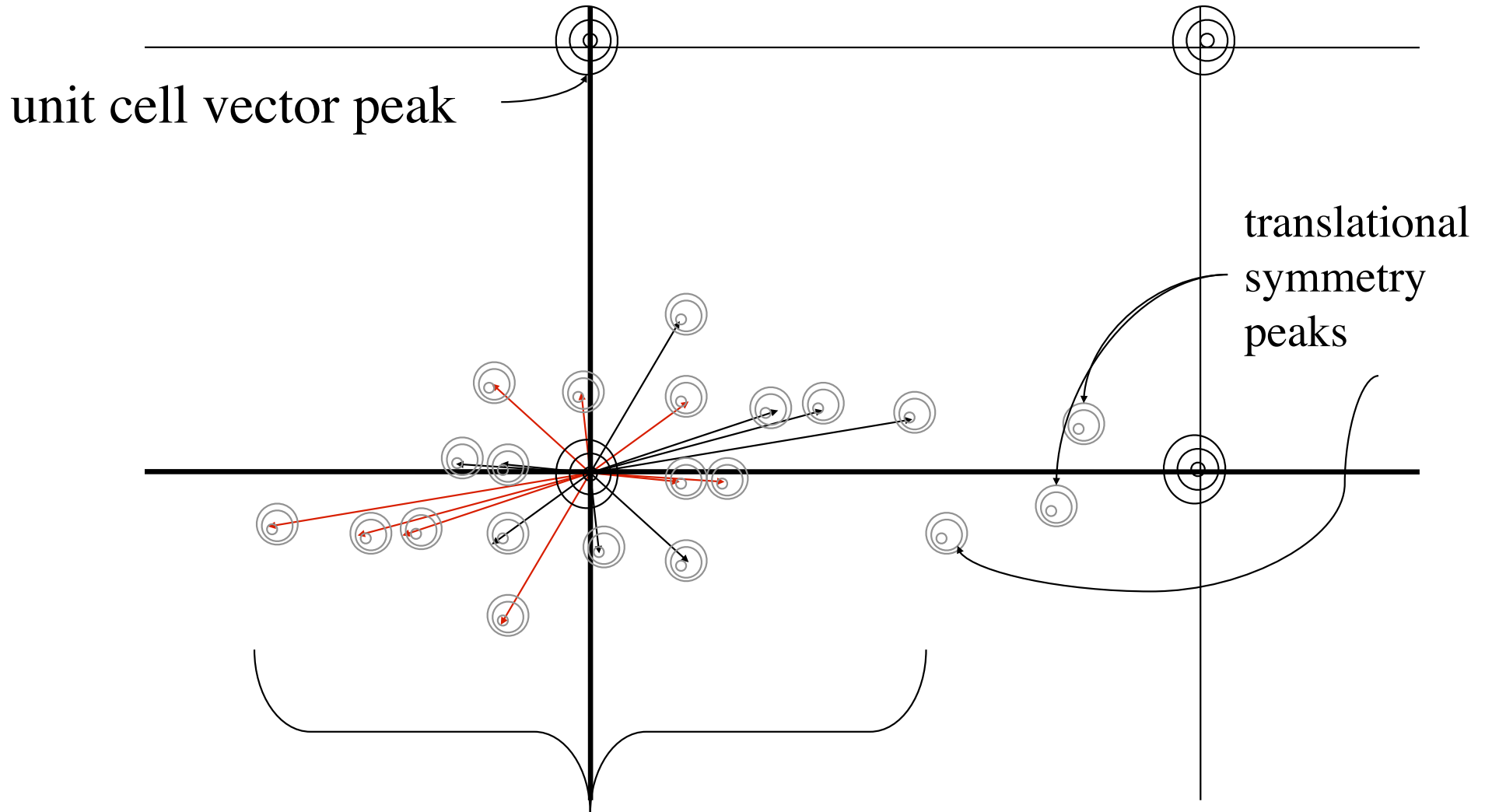
For example, take glycine, 5 atoms (not counting H's)



Move each vector to the origin

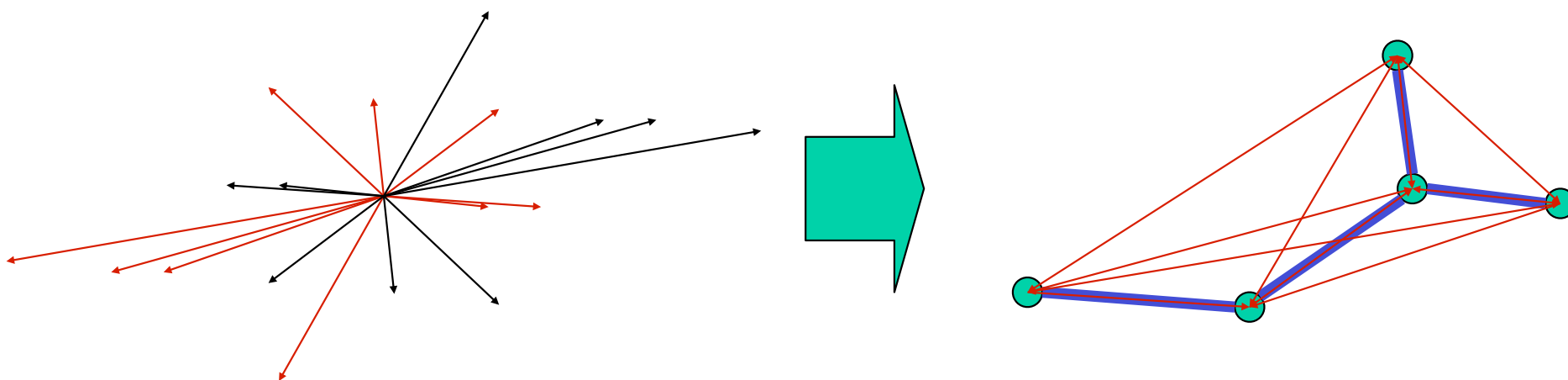


Pattern map for Gly in P1



Can you reassemble glycine from this?

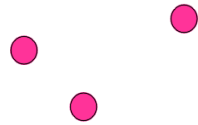
For small molecules, vector/geometry problem
can be solved...



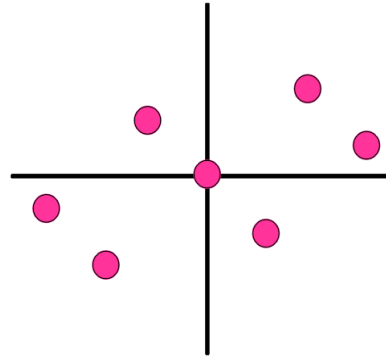
...if you know the stereochemistry (bond lengths, angles) of the molecule.

Rotation Function

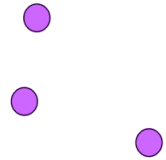
Target structure



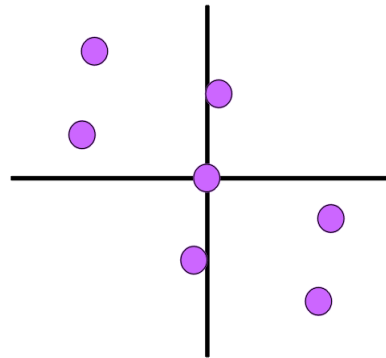
Target Patterson



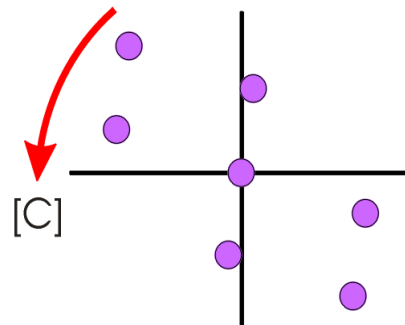
Search model



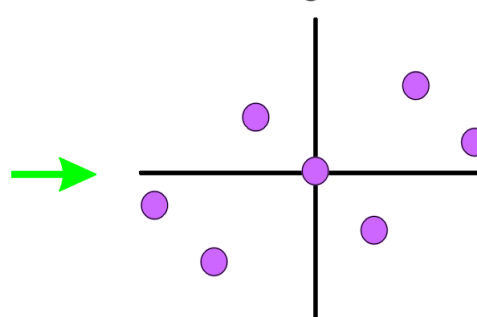
Search model Patterson



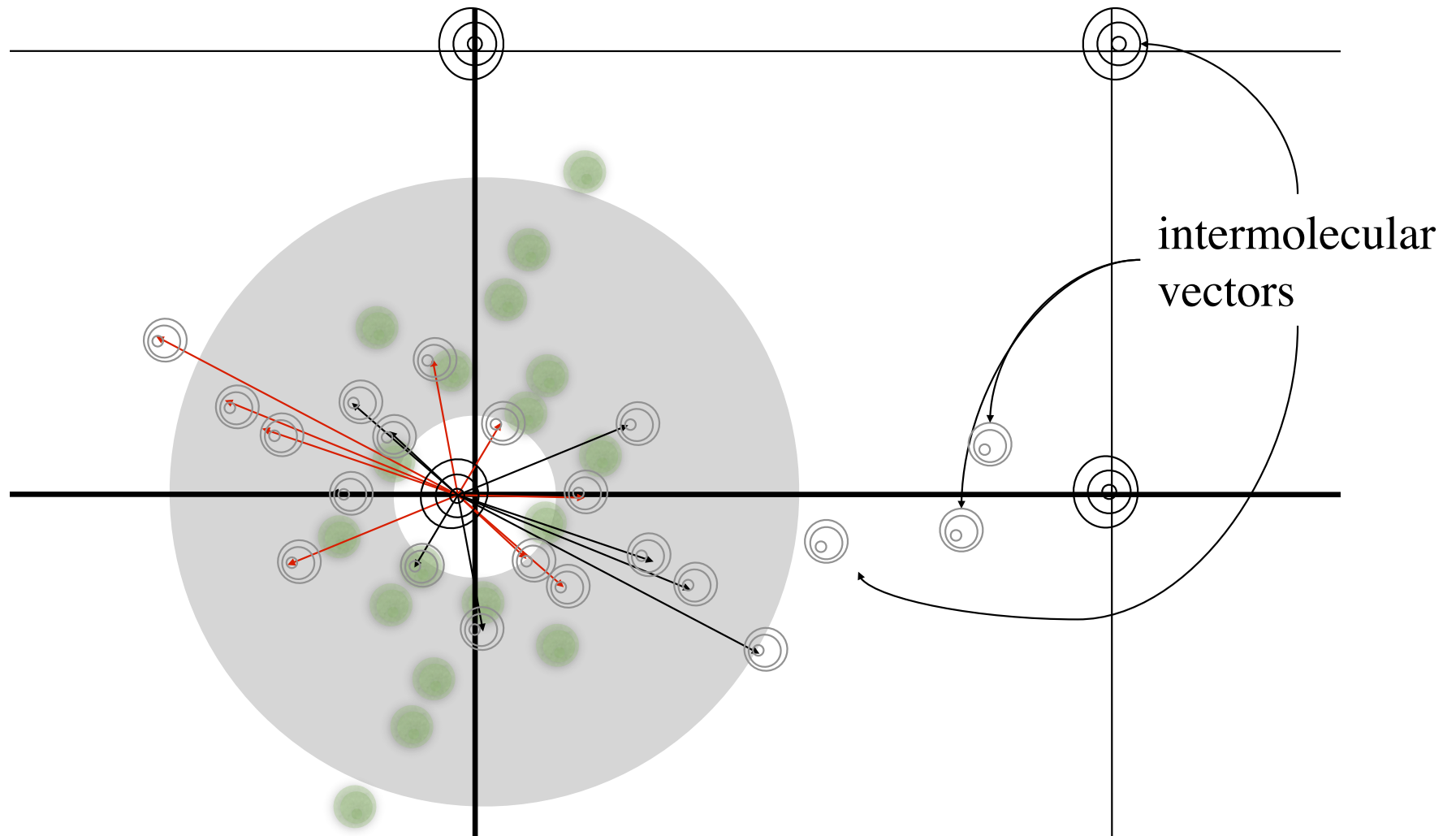
rotating Search model Patterson



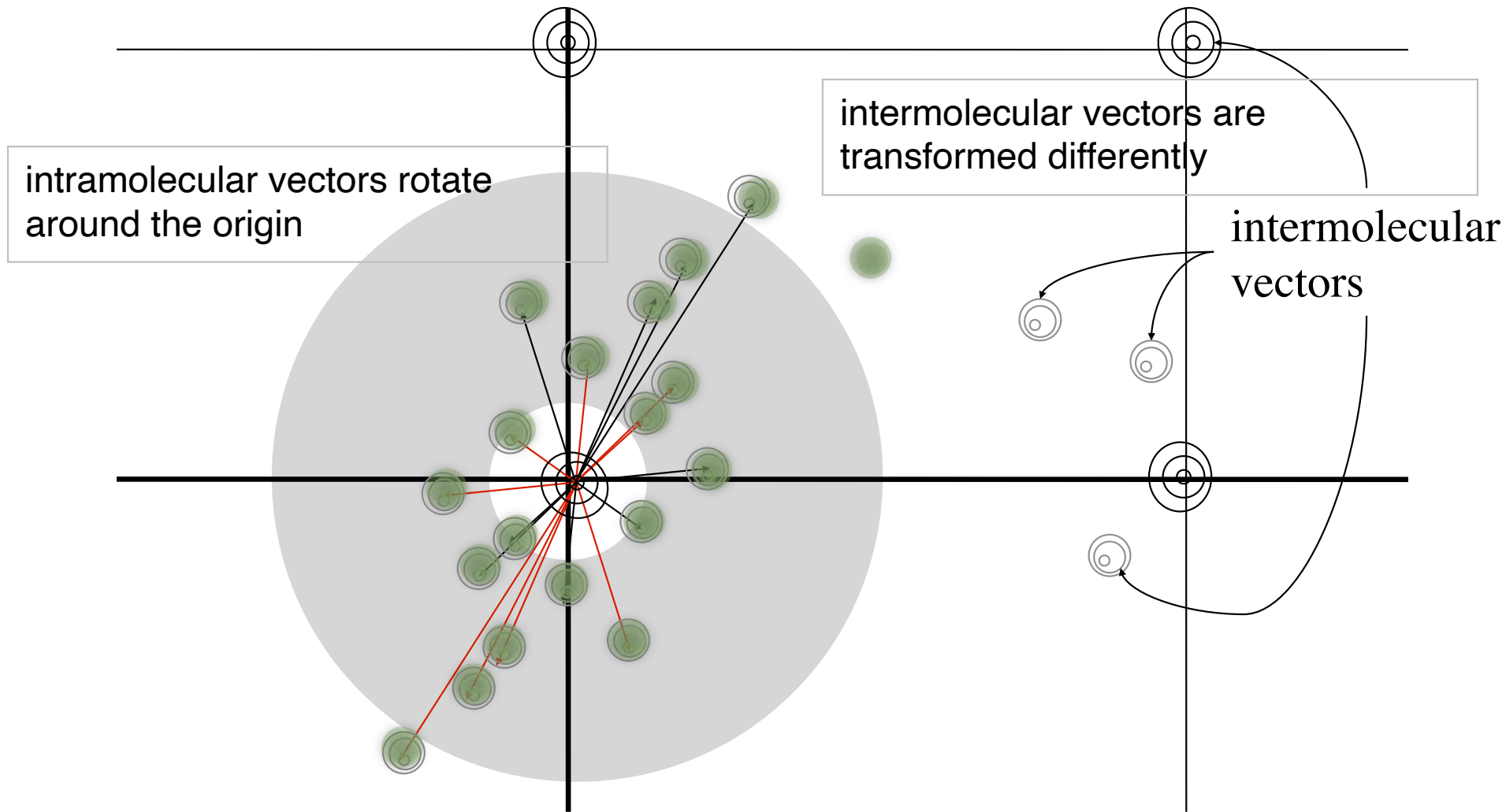
Matches target Patterson



Rotating the Patterson



Rotating the Patterson



The correlation function

The correlation between any two functions x and y is defined as:

$$r = \frac{\sum (x - \bar{x})(y - \bar{y})}{\sqrt{\sum (x - \bar{x})^2 \sum (y - \bar{y})^2}}$$

x and y are functions with the same range. \bar{x} means the mean value of the function x

If the correlation is perfect, $r=1.000$

If the anti-correlation is perfect, $r=-1.000$

If there is no correlation, r is close to zero.

Patterson correlation function

$$r = \frac{\sum (P_o(v) - \bar{P}_o)(P_{\text{mod}}(v) - \bar{P}_{\text{mod}})}{\sqrt{\sum (P_o(v) - \bar{P}_o)^2 \sum (P_{\text{mod}}(v) - \bar{P}_{\text{mod}})^2}}$$

= correlation between observed Patterson and rotated, model Patterson

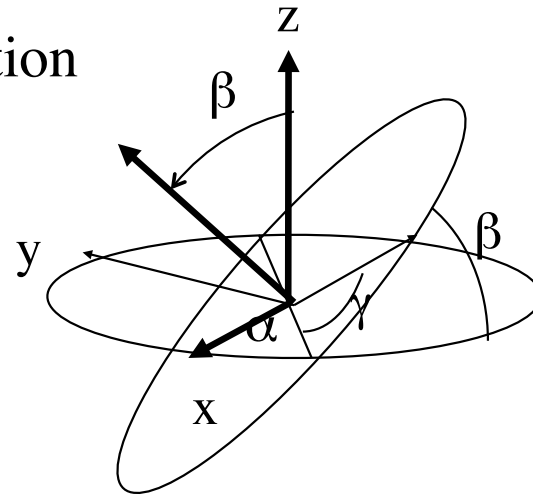
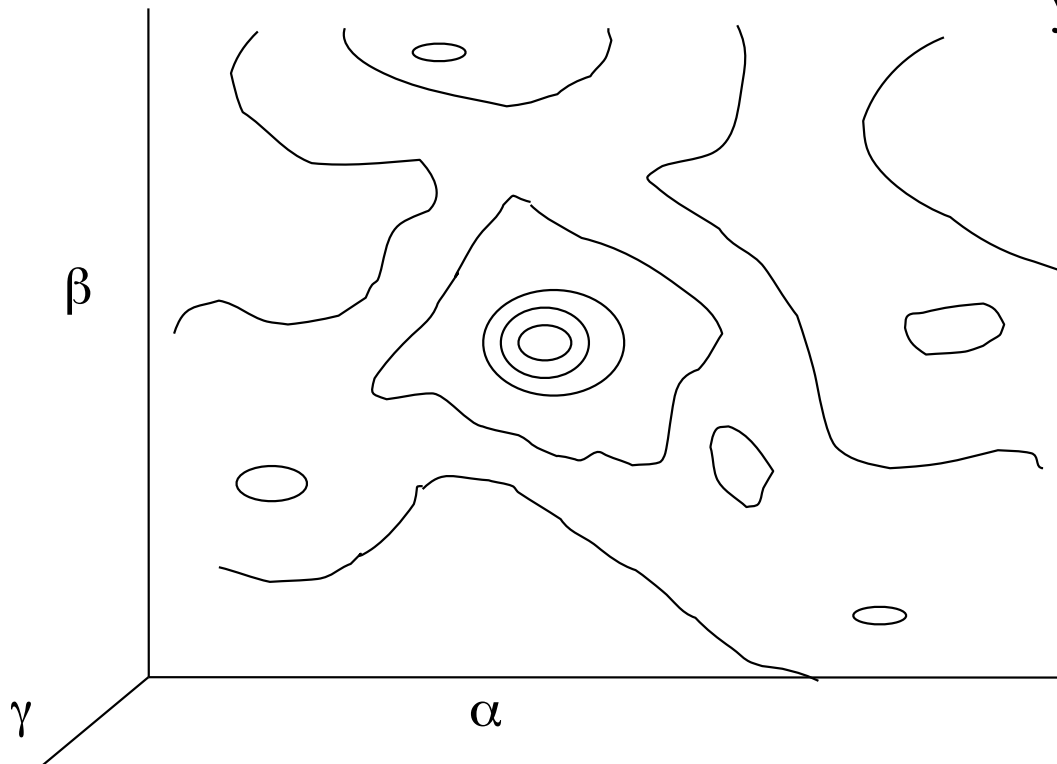
$(P_{\text{mod}}(v) - \bar{P}_{\text{mod}})$ = rotated model Patterson, at v , mean corrected.

The sums Σ are done over v in a spherical shell of the Patterson map that excludes the self-peak and the very long vectors.

Typically, $4\text{\AA} \leq |v| \leq 20\text{\AA}$, is a good range for the rotation function.

The Rotation Function

Three angles (α, β, γ) define all possible rigidbody rotations. The solution of the rotation function are the angles that give the highest *Patterson correlation function*.



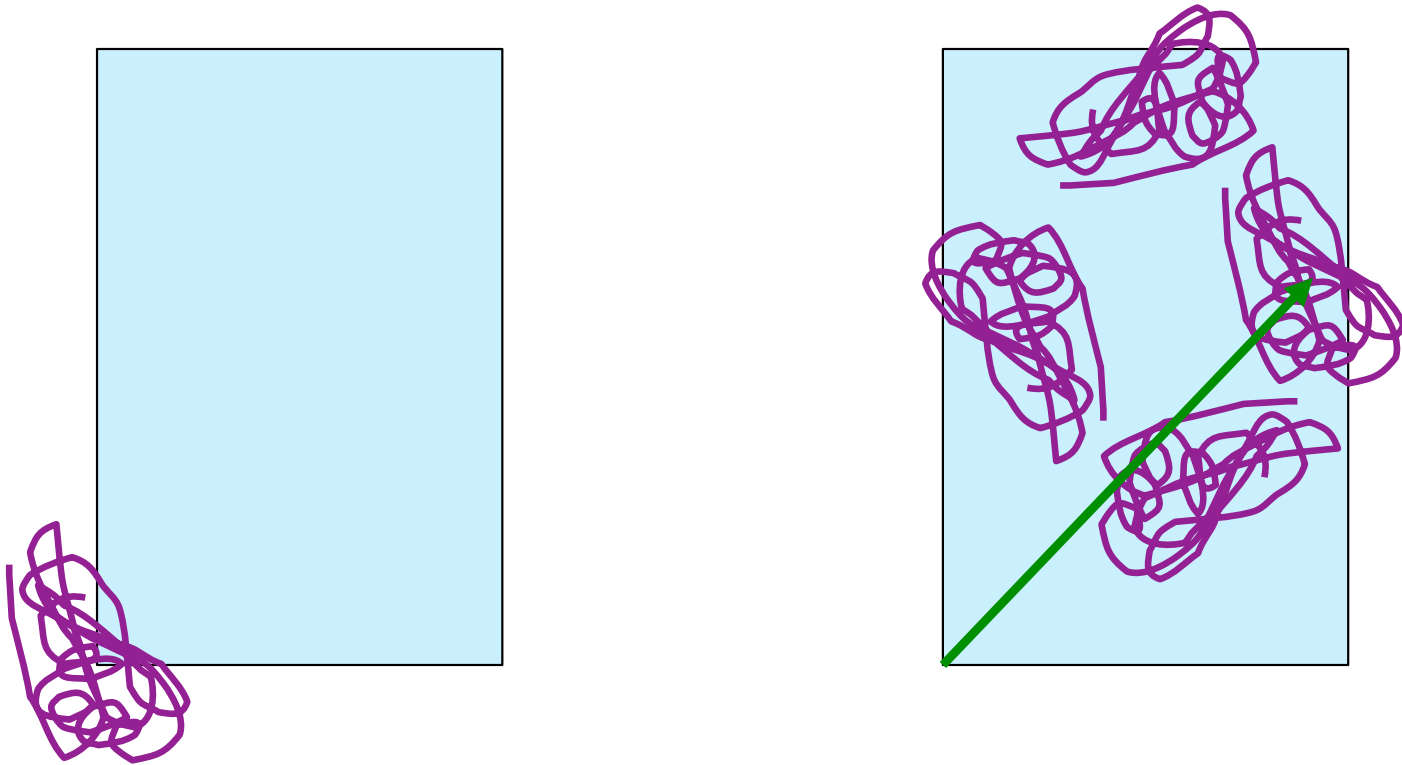
Non-crystallographic symmetry can be detected using the Self Rotation Function

If the native Patterson is rotated against itself and the correlation (r) is calculated, the result (call the “Self Rotation Function”) will have at a non-symmetry-related position only if the asymmetric unit has **NON-CRYSTALLOGRAPHIC SYMMETRY (NCS)**.

NCS means that an envelope of the asu exists for which:

$$\rho(r) = \rho(\underline{M}_{ncs}r + v_{ncs})$$

The Translation Function

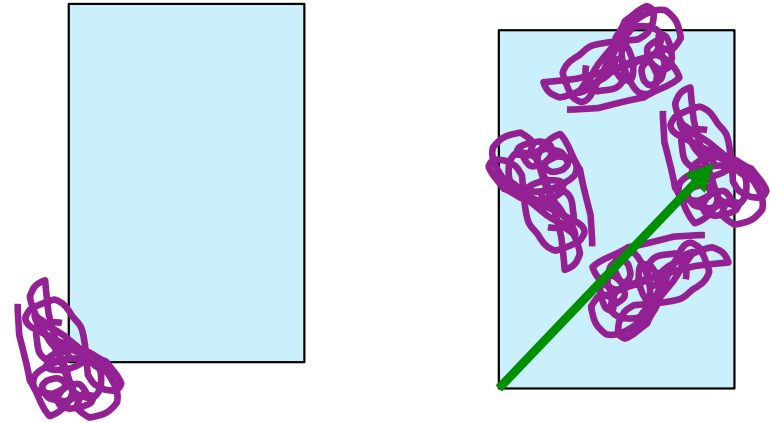


The model is oriented correctly with respect to the cell axes, but it is still at the origin. We need a translation vector (green) to translate the model to its position in the crystal unit cell relative to the origin.

How do we know which vector to use?

The Translation Function

1. Translate model
 2. Generate symmetry
 3. Check for good packing.
 4. Calculate R-factor
- ====> keep translation with best R-factor



A translation of the coordinates is:

$$x' = x + t$$

Symmetry positions are calculated using symops (M, v)

$$x' = Mx + v \quad (\text{M is the matrix and v is the vector})$$

Combining, we can express each translation as:

$$x' = M(x + t) + v = Mx + Mt + v$$

The Translation Function

Equation to calculate F_{calc} give
a translation vector t .

$$F_{\text{calc}}(\mathbf{h}) = \sum_{\underline{Z}} \sum_{\mathbf{g}} f_{\mathbf{g}}(\mathbf{h}) e^{i2\pi\mathbf{h}\cdot(\underline{Z}(\mathbf{r}_{\mathbf{g}}+\mathbf{t}))}$$

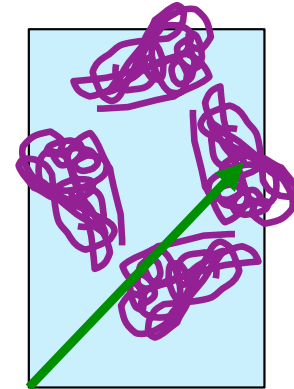
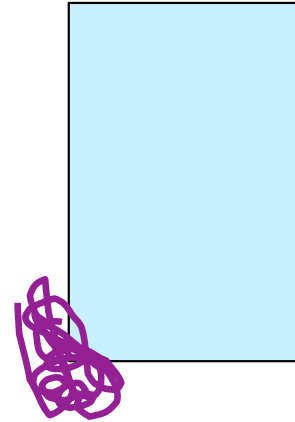
where

\mathbf{g} = all atoms,

\underline{Z} = all symops,

$\mathbf{r}_{\mathbf{g}}$ = fractional coordinates of atom \mathbf{g} ,

\mathbf{t} = translation vector.



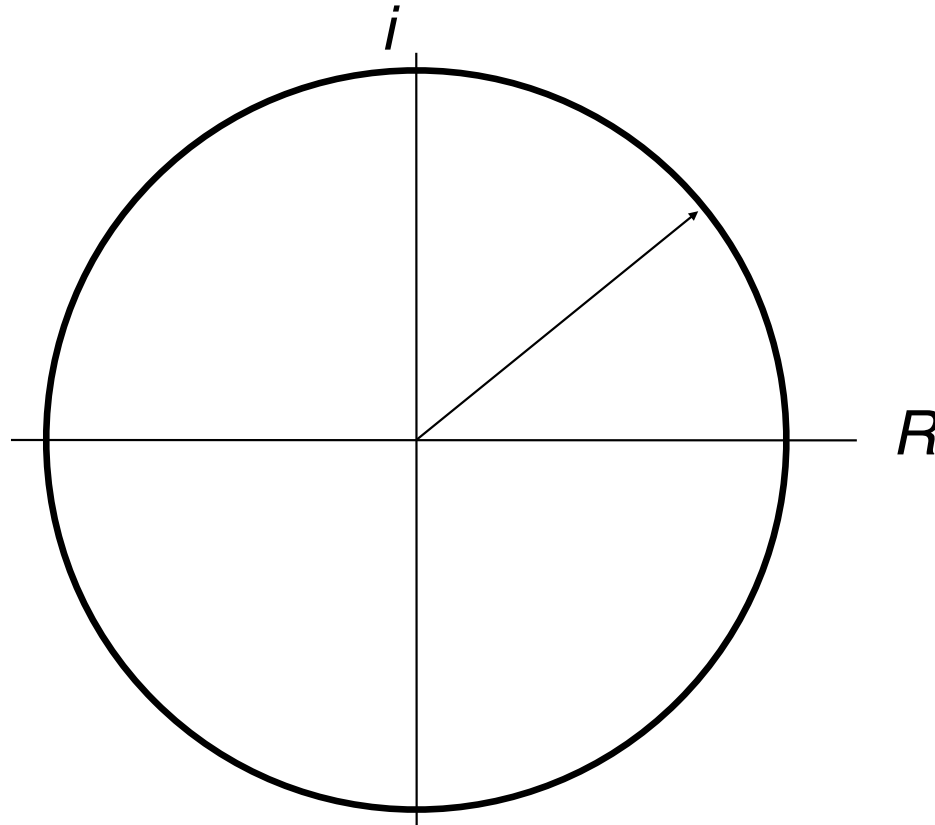
Equation to calculate R-factor given F_{calc} as a function of \mathbf{t} .

$$R(t) = \frac{\sum_h \left| |F_{\text{obs}}(h)| - k |F_{\text{calc}}(h, t)| \right|}{\sum_h |F_{\text{obs}}(h)|}$$

Heavy Atom Methods

<div style="display: flex; justify-content: space-around; align-items: center;"> <div style="background-color: #d3d3d3; padding: 5px;">Non-metal</div> <div style="background-color: #90ee90; padding: 5px;">Metal</div> <div style="background-color: #add8e6; padding: 5px;">Noble gas</div> </div> <div style="display: flex; justify-content: space-around; align-items: center; margin-top: 5px;"> <div style="background-color: #ffcc99; padding: 5px;">Alkali metal</div> <div style="background-color: #90ee90; padding: 5px;">Metalloid</div> <div style="background-color: #ffcc99; padding: 5px;">Lanthanide</div> </div> <div style="display: flex; justify-content: space-around; align-items: center; margin-top: 5px;"> <div style="background-color: #ff9966; padding: 5px;">Alkaline earth metal</div> <div style="background-color: #add8e6; padding: 5px;">Halogen</div> <div style="background-color: #cccccc; padding: 5px;">Actinide</div> </div> <div style="display: flex; justify-content: space-around; align-items: center; margin-top: 5px;"> <div style="background-color: #cccccc; padding: 5px;">Transition metal</div> </div>																							
1 H HYDROGEN 1.0079																	2 He HELIUM 4.0026						
3 Li LITHIUM 6.941	4 Be BERYLLIUM 9.0122																	5 B BORON 10.811	6 C CARBON 12.011	7 N NITROGEN 14.007	8 O OXYGEN 15.999	9 F FLUORINE 18.998	10 Ne NEON 20.1797
11 Na SODIUM 22.989	12 Mg MAGNESIUM 24.305																	13 Al ALUMINIUM 26.981	14 Si SILICON 28.085	15 P PHOSPHORUS 30.974	16 S SULFUR 32.066	17 Cl CHLORINE 35.453	18 Ar ARGON 39.948
19 K POTASSIUM 39.098	20 Ca CALCIUM 40.078	21 Sc SCANDIUM 44.955	22 Ti TITANIUM 47.867	23 V VANADIUM 50.9415	24 Cr CHROMIUM 51.9961	25 Mn MANGANESE 54.938	26 Fe IRON 55.845	27 Co COBALT 58.933	28 Ni NICKEL 58.6934	29 Cu COPPER 63.546	30 Zn ZINC 65.38	31 Ga GALLIUM 69.723	32 Ge GERMANIUM 72.63	33 As ARSENIC 74.921	34 Se SELENIUM 78.971	35 Br BROMINE 79.904	36 Kr KRYPTON 83.798						
37 Rb RUBIDIUM 85.467	38 Sr STRONTIUM 87.62	39 Y YTTORIUM 88.9058	40 Zr ZIRCONIUM 91.224	41 Nb NIOBIUM 92.9063	42 Mo MOLYBDENUM 95.95	43 Tc TECHNETIUM (98)	44 Ru RUTHENIUM 101.07	45 Rh RHODIUM 102.90	46 Pd PALLADIUM 106.42	47 Ag SILVER 107.8682	48 Cd CADMIUM 112.414	49 In INDIUM 114.818	50 Sn TIN 118.710	51 Sb ANTIMONY 121.760	52 Te TELLURIUM 127.60	53 I IODINE 126.90	54 Xe XENON 131.293						
55 Cs CAESIUM 132.905	56 Ba BARIUM 137.327	57-71*	72 Hf HAFNIUM 178.49	73 Ta TANTALUM 180.94	74 W TUNGSTEN 183.84	75 Re RHENIUM 186.207	76 Os OSMIUM 190.23	77 Ir IRIDIUM 192.217	78 Pt PLATINUM 195.084	79 Au GOLD 196.96	80 Hg MERCURY 200.59	81 Tl THALLIUM 204.38	82 Pb LEAD 207.2	83 Bi BISMUTH 208.98	84 Po POLONIUM (209)	85 At ASTATINE (210)	86 Rn RADON (222)						
87 Fr FRANCIUM (223)	88 Ra RADIUM (226)	89-103**	104 Rf RUTHERFORDIUM (267)	105 Db DUBNIUM (268)	106 Sg SEABORGIUM (271)	107 Bh BOHRIUM (272)	108 Hs HASSIUM (270)	109 Mt MEITNERIUM (276)	110 Ds DARMSTADIUM (281)	111 Rg ROENTGENIUM (280)	112 Cn COPERNICIUM (285)	113 Uut UNUNTRIUM (284)	114 Fl FLEROVIUM (289)	115 Uup UNUNPENTIUM (288)	116 Lv LIVERMORIUM (293)	117 Uus UNUNSEPTIUM (294)	118 Uuo UNUNOCTIUM (294)						
			57 La LANTHANUM 138.90	58 Ce CERIUM 140.116	59 Pr PRASEODYMIUM 140.90	60 Nd NEODYMIUM 144.242	61 Pm PROMETHIUM (145)	62 Sm SAMARIUM 150.36	63 Eu EUROPIUM 151.964	64 Gd GADOLINIUM 157.25	65 Tb TERBIUM 158.92	66 Dy DYSPROSIUM 162.500	67 Ho HOLMIUM 164.93	68 Er ERBIUM 167.259	69 Tm THULIUM 168.93	70 Yb YTTERIUM 173.054	71 Lu LUTETIUM 174.9668						
			89 Ac ACTINIUM (227)	90 Th THORIUM 232.0377	91 Pa PROTACTINIUM 231.03	92 U URANIUM 238.02	93 Np NEPTUNIUM (237)	94 Pu PLUTONIUM (244)	95 Am AMERICIUM (243)	96 Cm CURIUM (247)	97 Bk BERKELIUM (247)	98 Cf CALIFORNIUM (251)	99 Es EINSTEINIUM (252)	100 Fm FERMIUM (257)	101 Md MENDELEVIUM (258)	102 No NOBELIUM (259)	103 Lr LAWRENCIUM (262)						

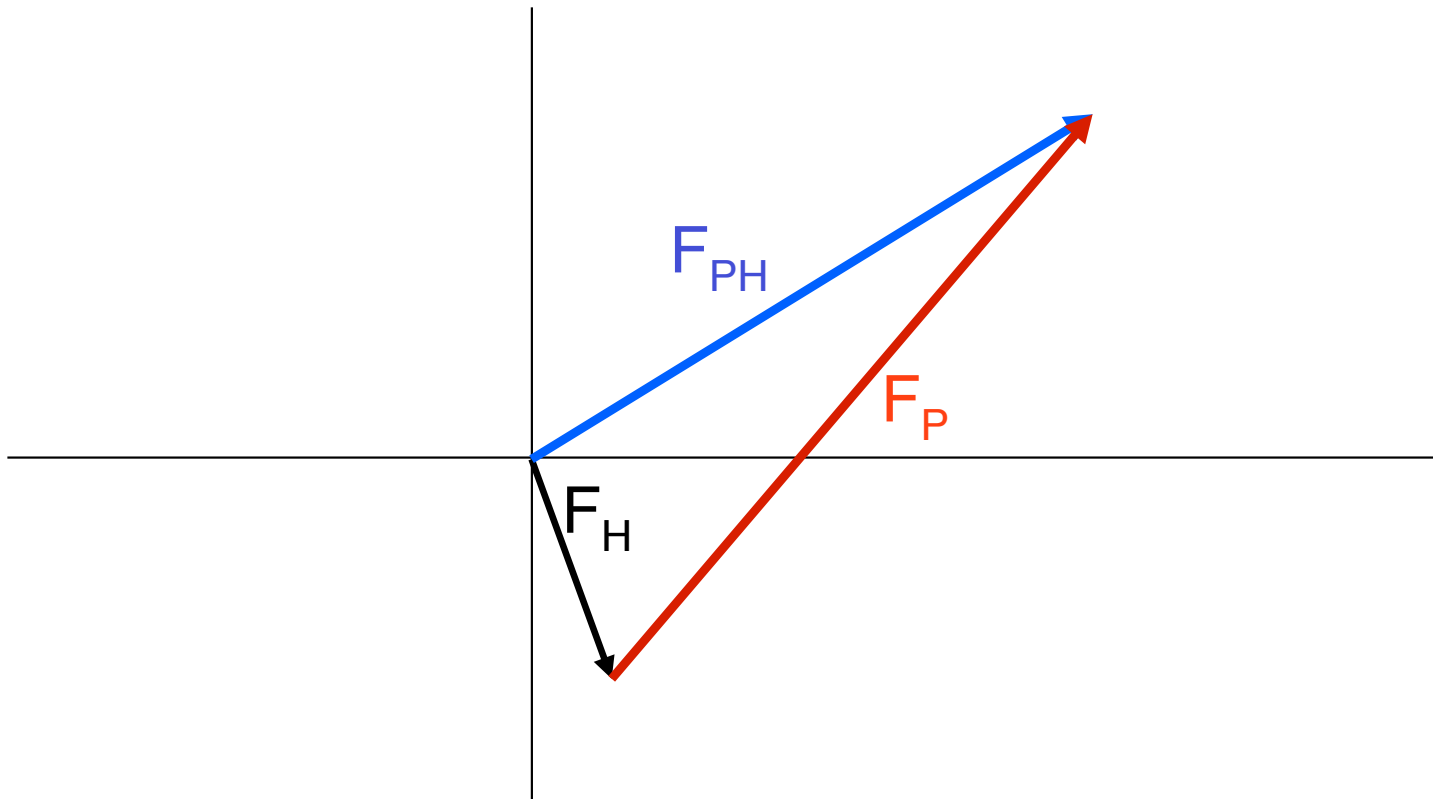
We can represent a structure factor of known amplitude and unknown phase as a circle in Argand space.



Radius of the circle is the amplitude. The true F lies somewhere on the circle.

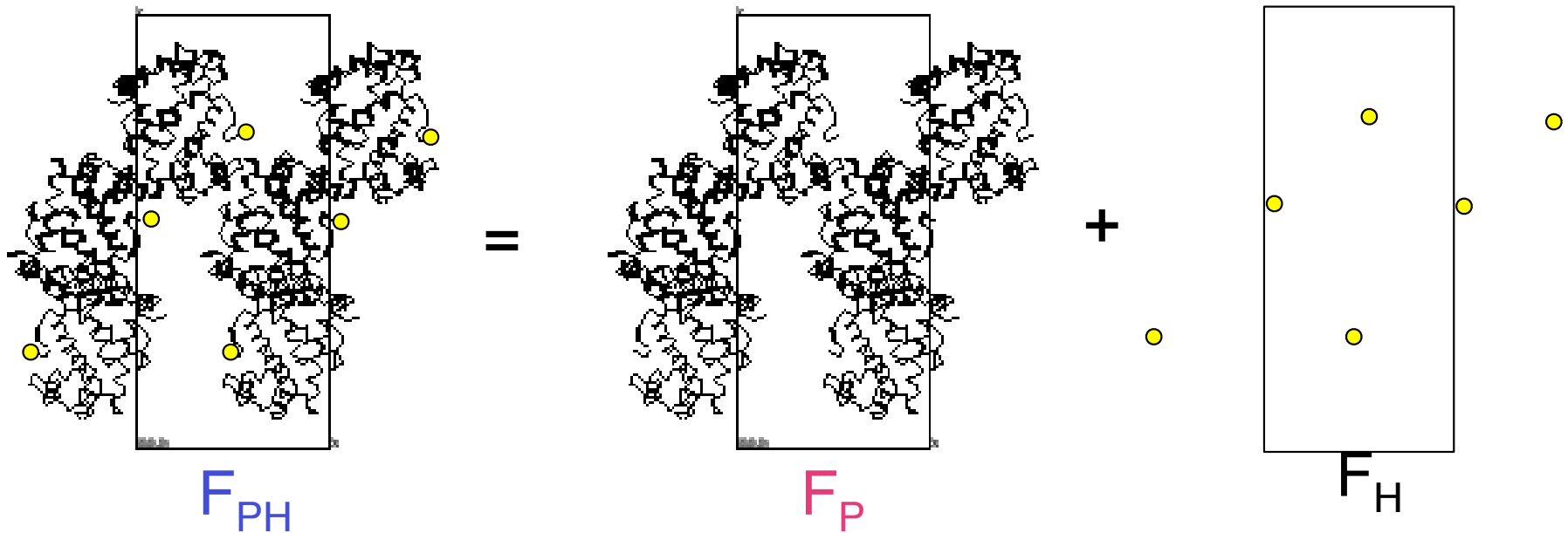
Heavy atom isomorphous replacement

$$F_P + F_H = F_{PH}$$



Heavy atom isomorphous replacement

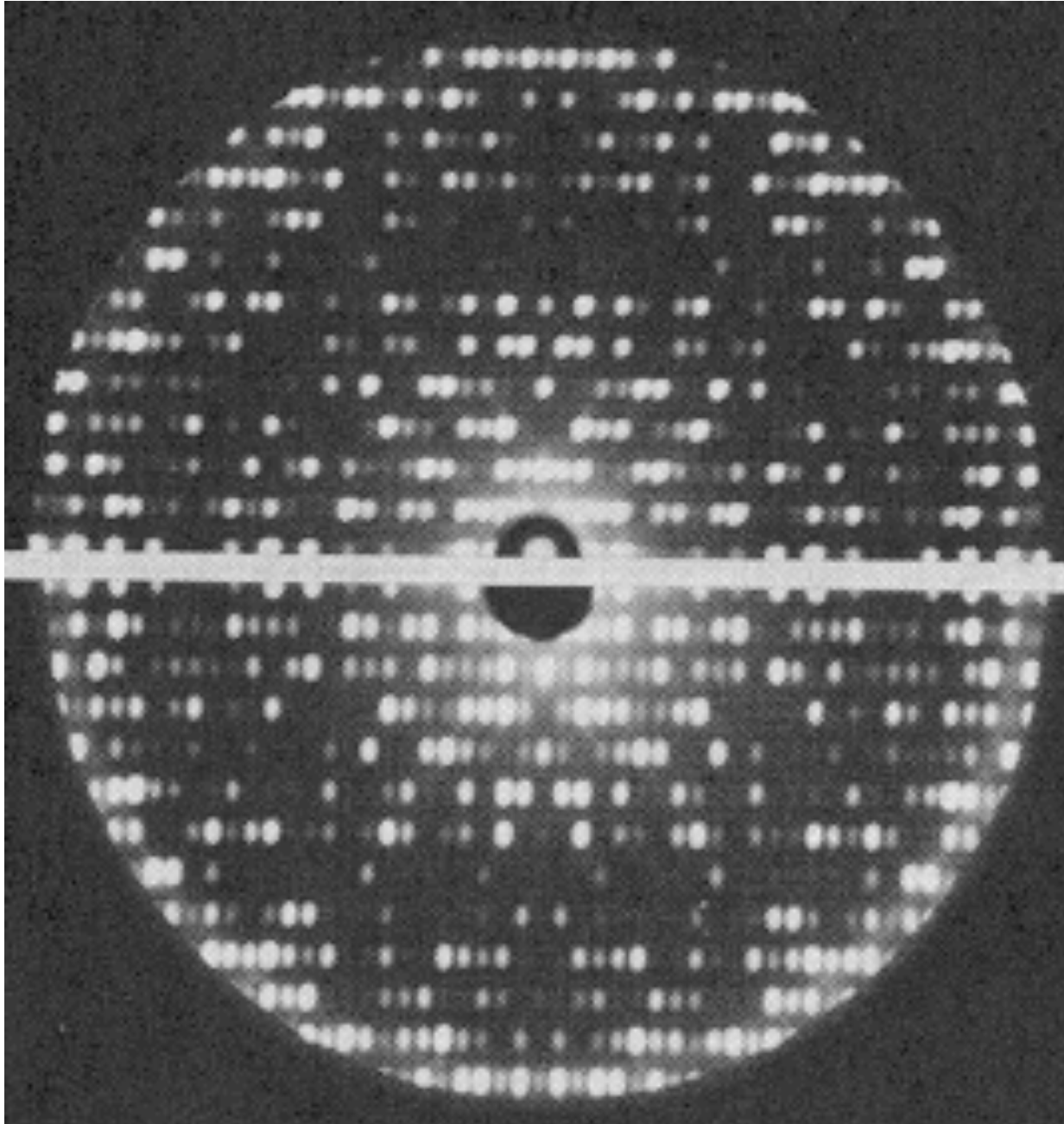
=Turning proteins into small molecules by soaking in heavy atoms



The Fourier transform (i.e. diffraction pattern) of a heavy atom derivative is the vector sum of the transforms of the protein and the heavy atoms.

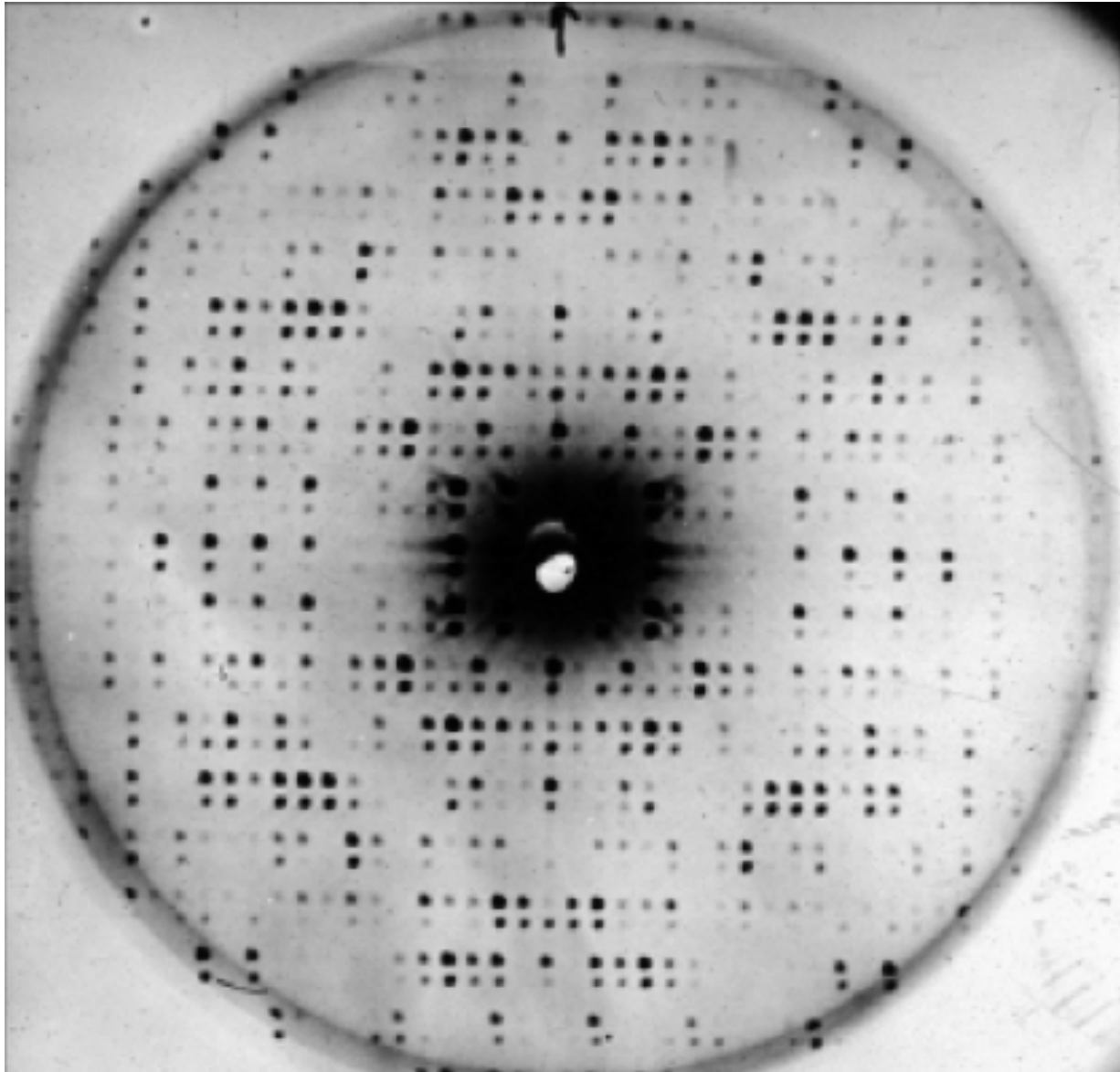
NOTE: protein and protein-heavy atom crystals must be *isomorphous*.

Comparing parent and heavy atom data sets.

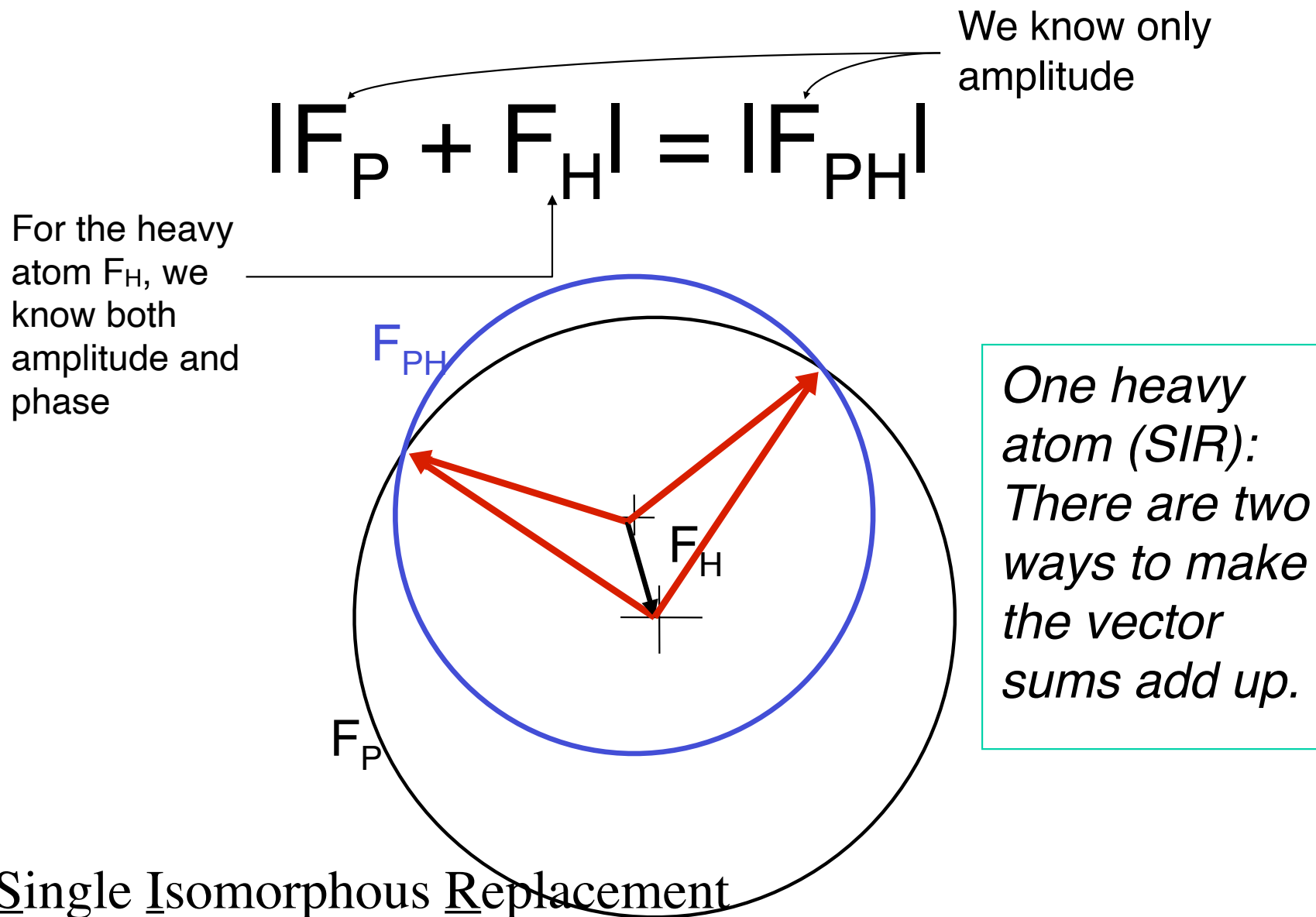


The upper and lower images are two precession photographs, showing the $l=0$ level of reciprocal space. The upper is the original protein crystal. The lower is after soaking in heavy atom solution. Note changes in intensity.

Two protein diffraction patterns superimposed and shifted vertically relative to one another. One is from native bovine β -lactoglobulin and the other is from a crystal soaked in a mercury-salt solution. Note the intensity changes for certain reflections and the identical unit cells (spacing of the spots) suggesting isomorphism. (Photograph courtesy of Professor Lindsay Sawyer.)

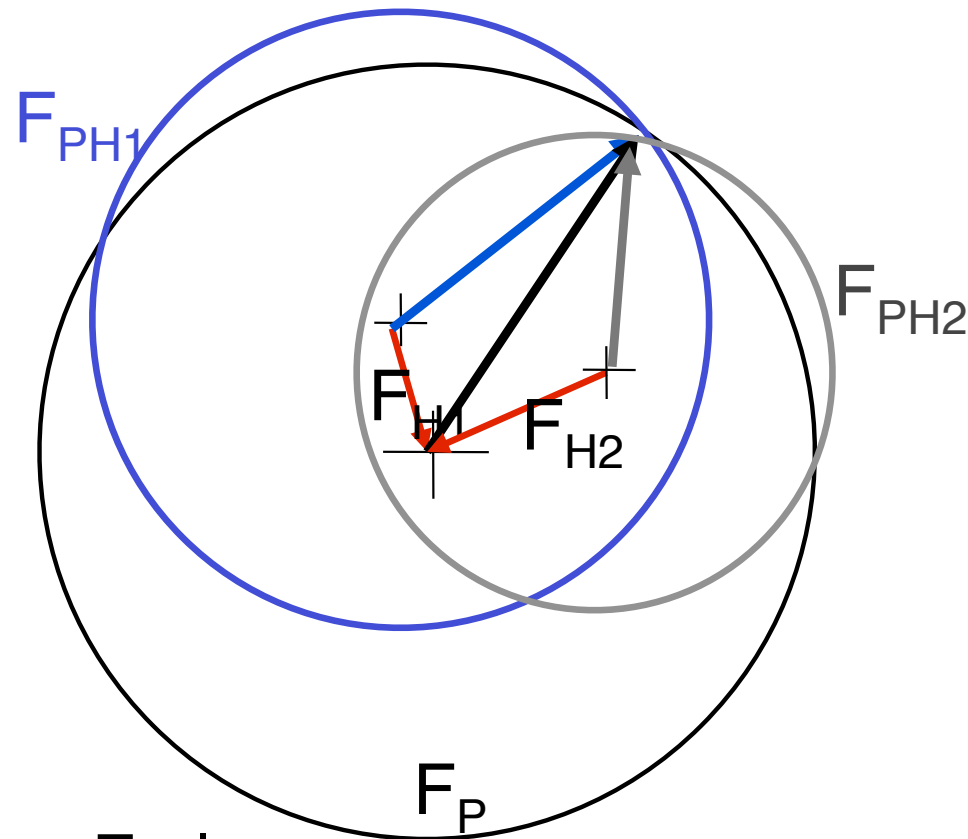


Harker diagram method for discovering phase from amplitudes.



Two heavy atom derivatives (MIR), produce unambiguous phases

Multiple Isomorphous Replacement



$$|F_P + F_{H1}| = |F_{PH1}|$$

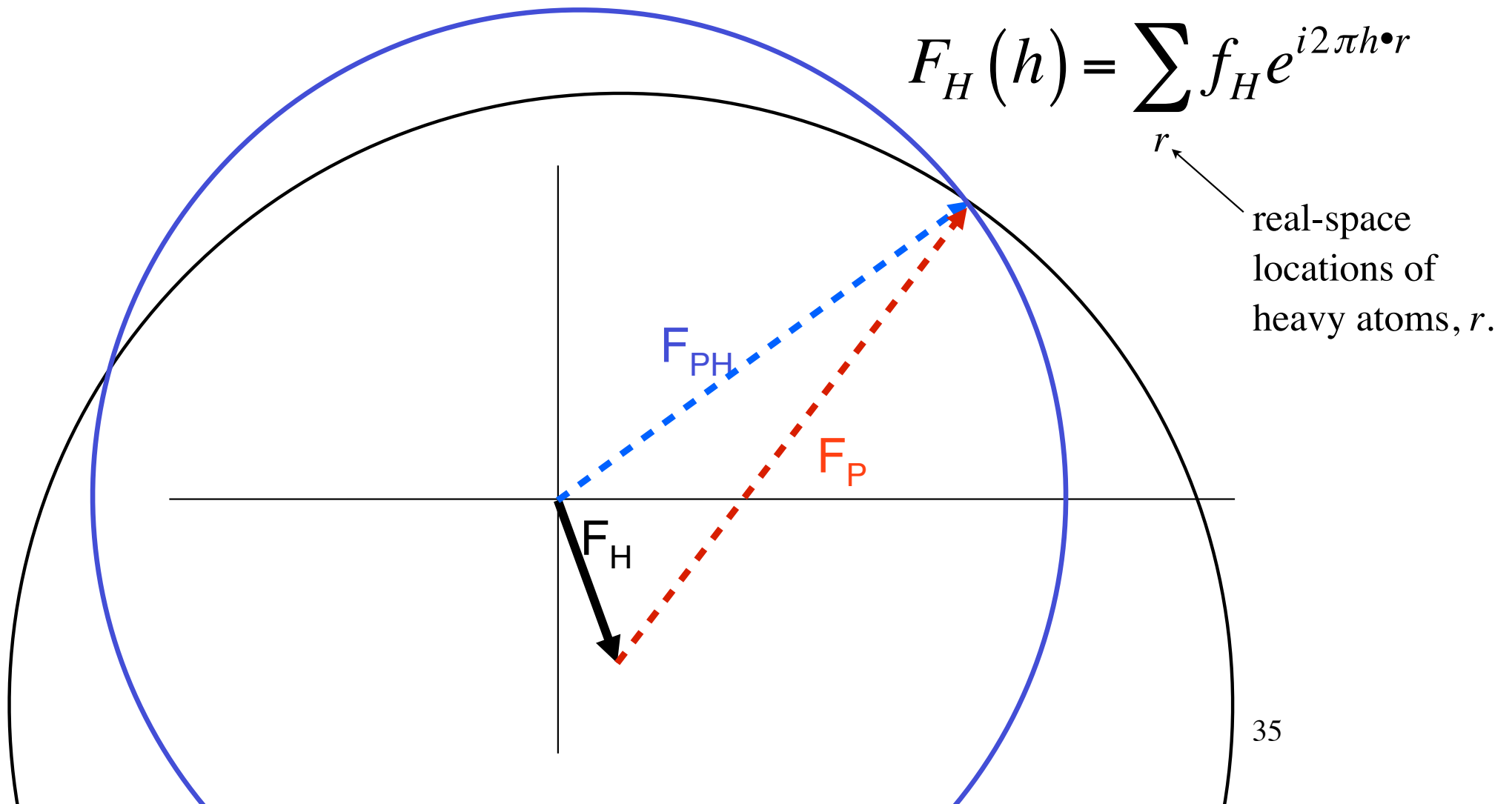
$$|F_P + F_{H2}| = |F_{PH2}|$$

Or

$$|F_P| = |F_{PH1} - F_{H1}| = |F_{PH2} - F_{H2}|$$

- Phases are more important than amplitudes.
- We can't measure phases, only amplitudes.
- By adding heavy atoms, we change the amplitudes by a significant amount.
- If we know the contribution of the heavy atom, we can solve for the phase of the protein.
- SIR = single isomorphous replacement, gives an ambiguous phase.
- MIR = multiple isomorphous replacement, gives an unambiguous answer.

If we can locate the heavy atom, then we can calculate its contribution to F_{PH}

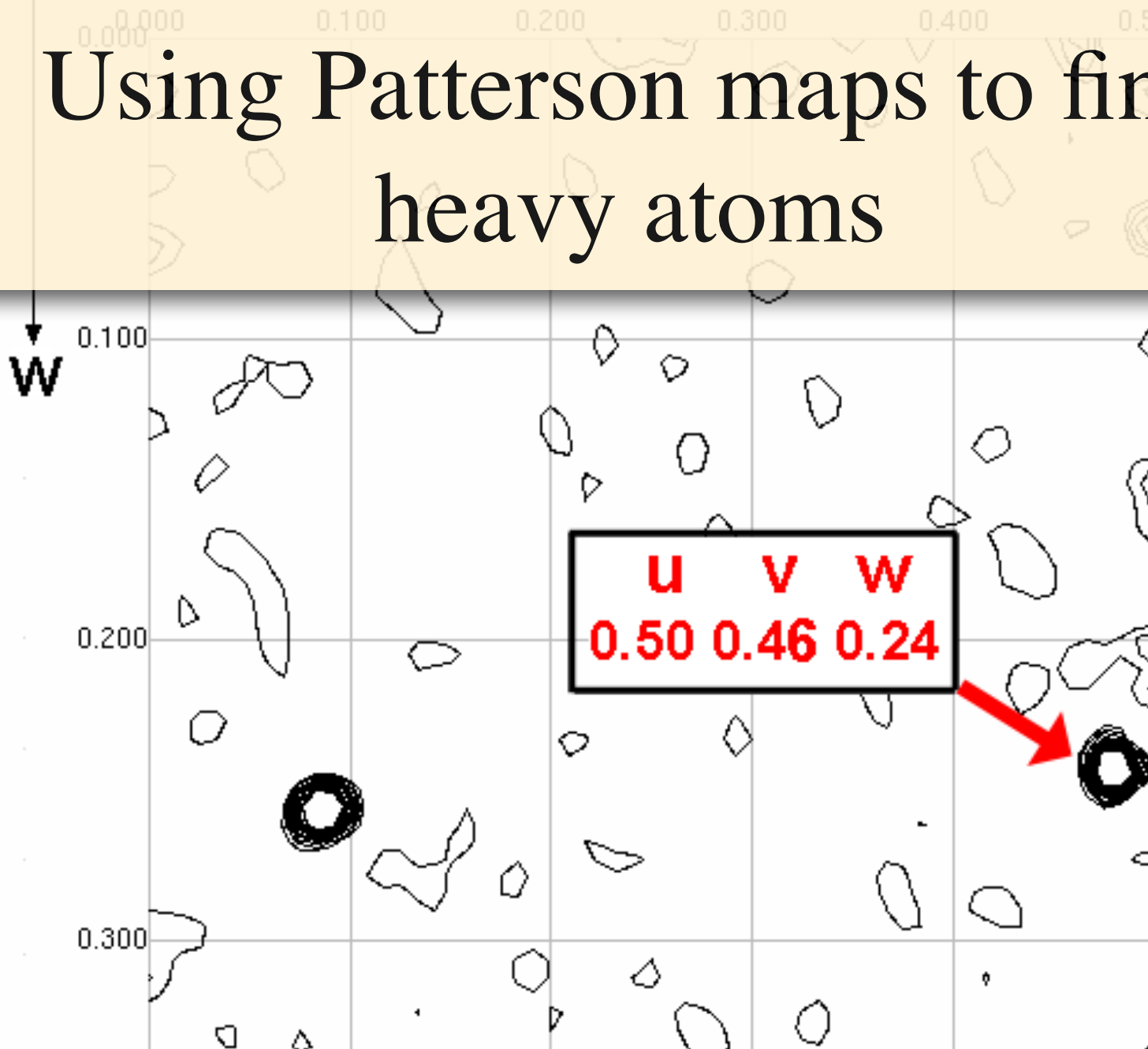


isomorphous difference Patterson map

$u=0.50$

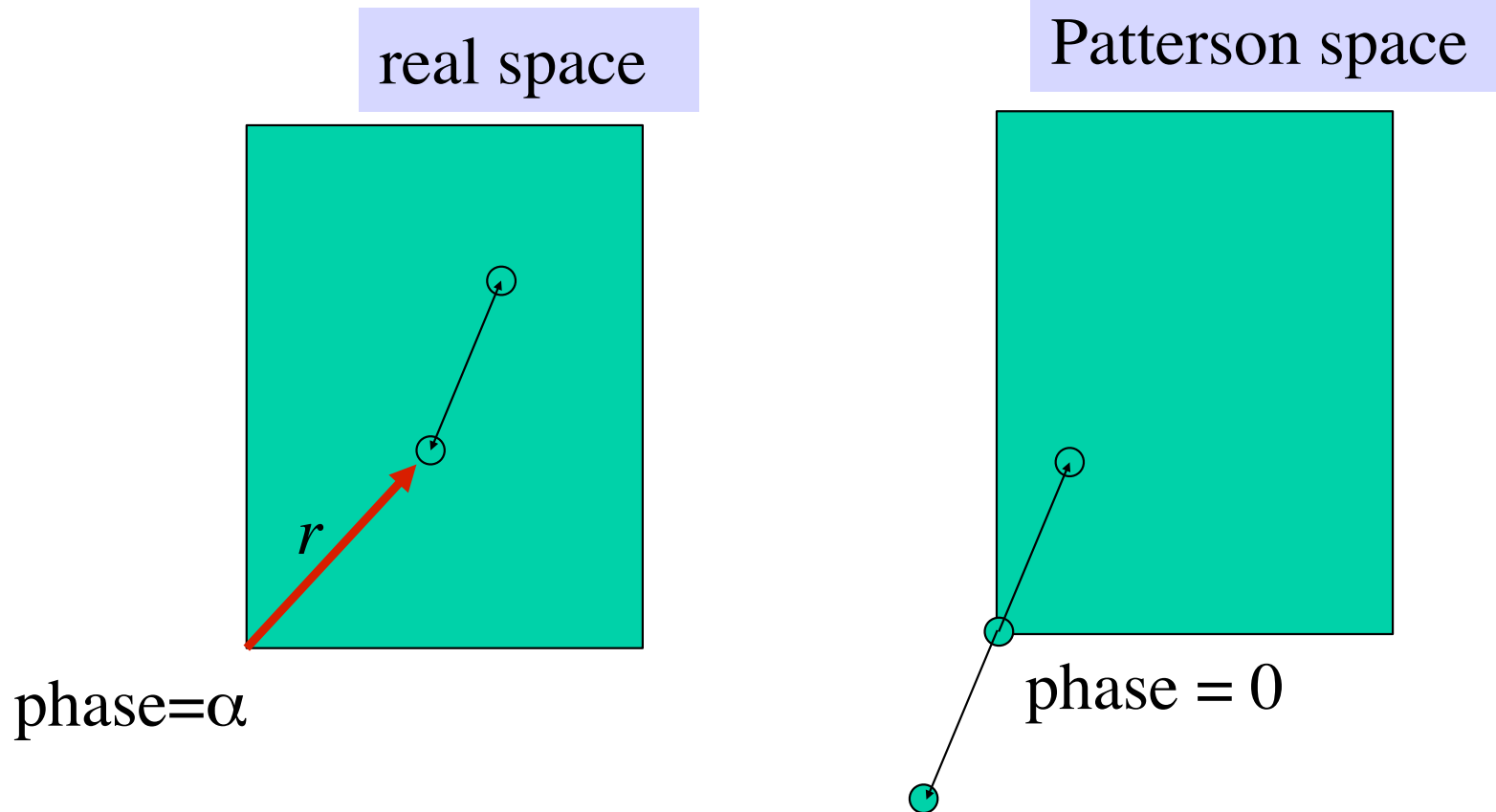
V

Using Patterson maps to find heavy atoms



“Patterson space”

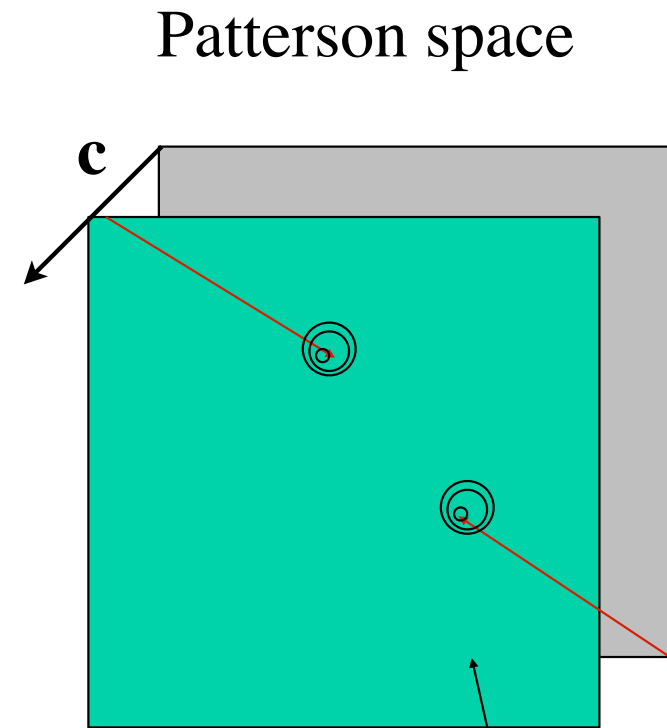
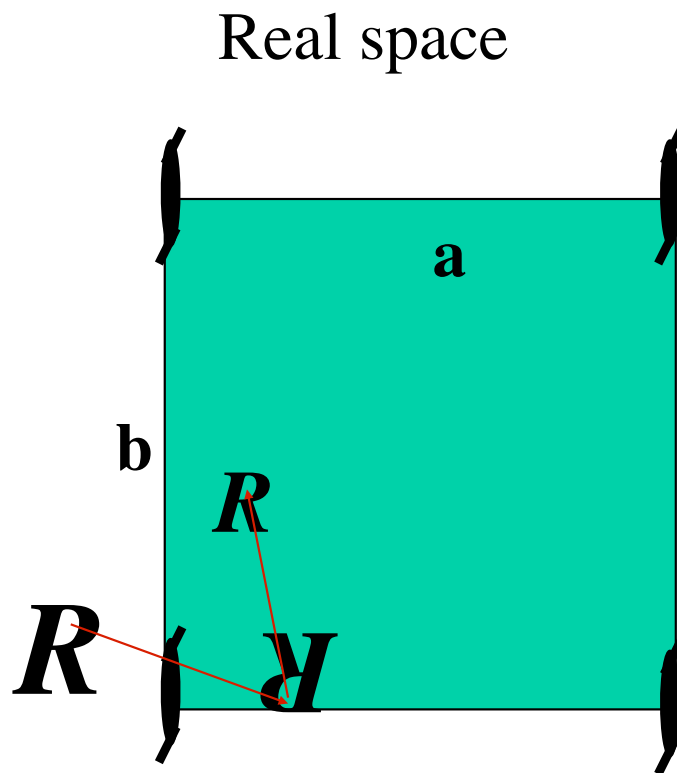
The Patterson map is the centrosymmetric projection



Using observed amplitudes, but setting all phases to 0 creates a centro-symmetric image of the molecule.

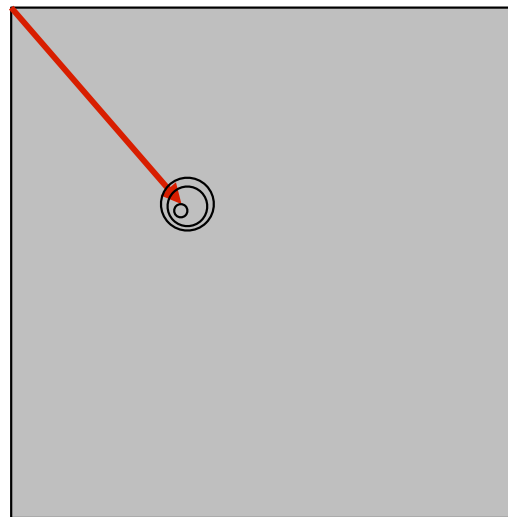
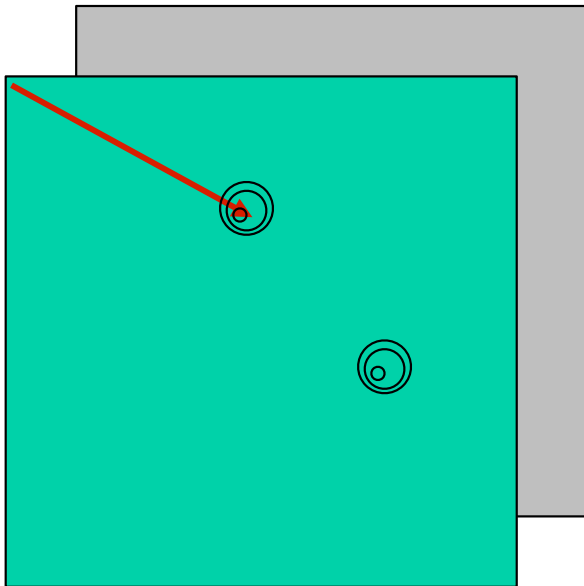
Patterson peaks generated by symmetry operations found are on *Harker sections*

$P2_1$

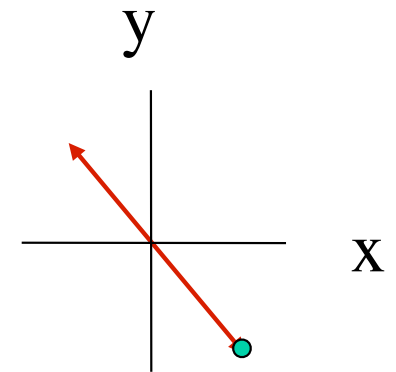


Harker section $z=0.5$

Harker sections tell us the location of atoms relative to the cell axes



If operator is a 2-fold, divide this vector by two to get the XY coordinates.



(The Z position is found on other sections)

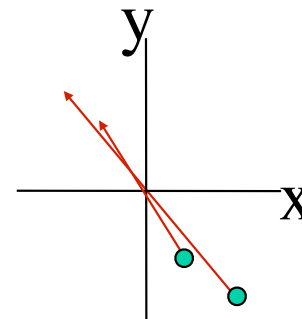
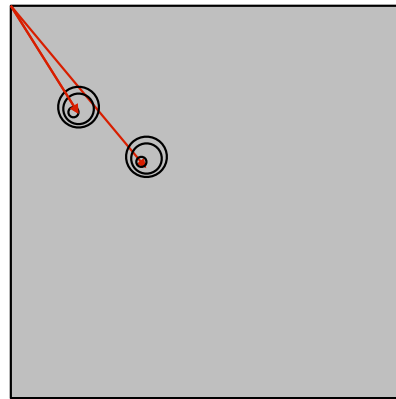
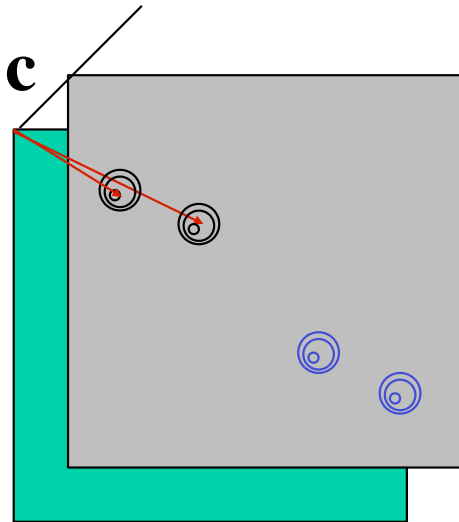
Non-Harker sections tell us inter-atomic vectors not related by symmetry

If there is more than one atom in the *asu*, you can get the vector between them by searching for peaks in non-Harker sections of the Patterson. (like the glycine example)

Then, combining knowledge from Harker sections (giving absolute positions) and non-Harker sections (giving relative positions) we can get the atomic coordinates.

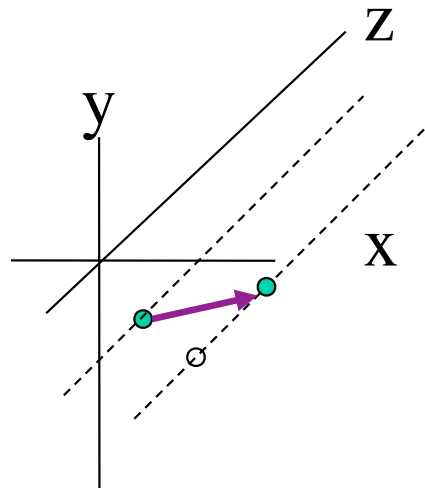
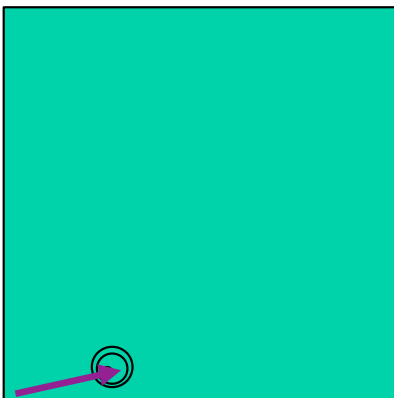
Simple case: 2 atoms, $P2_1$

In a Harker section



The xy-position is found relative to the 2-fold axis, for each atom

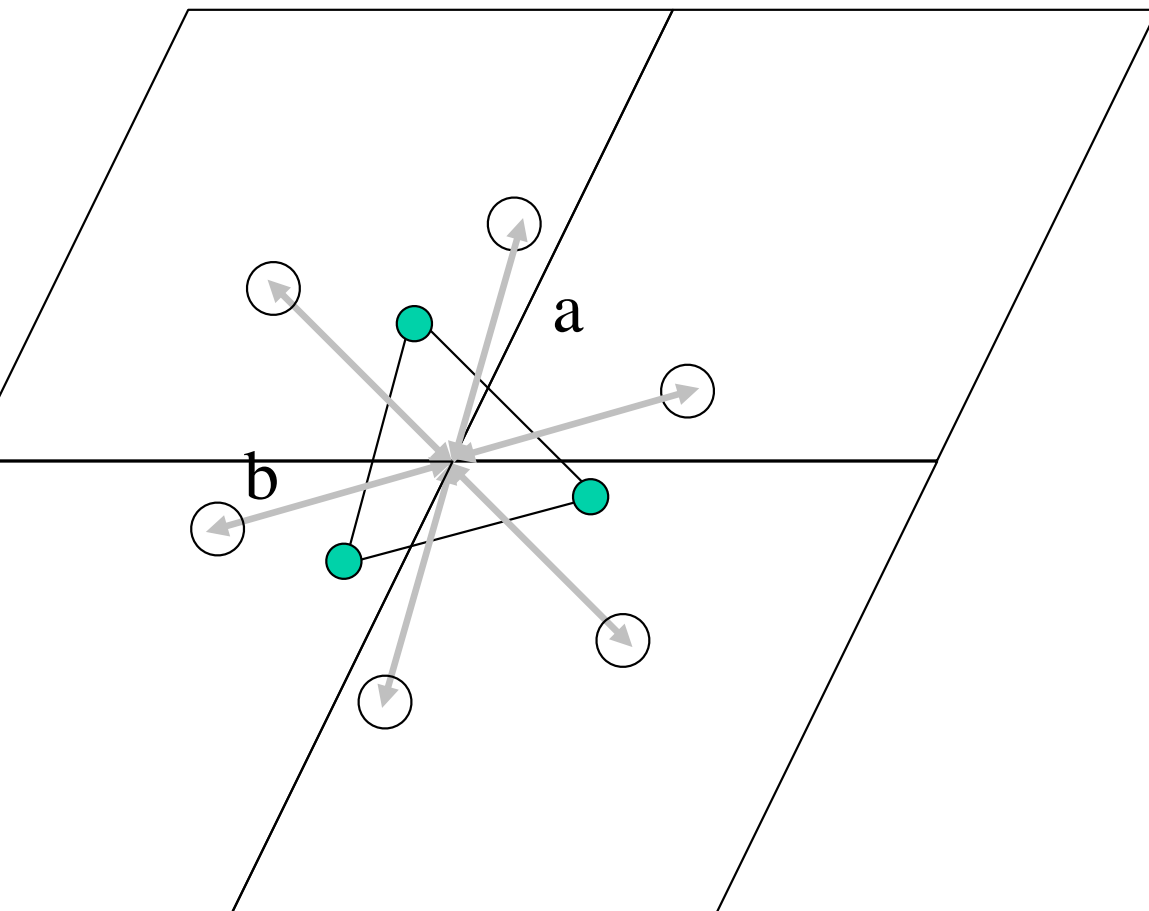
In a non-Harker section



The relative Z position is found for one atom relative to the other.

- A Patterson map can be calculated without needing to know the phases.
- A Patterson map shows the vectors between heavy atoms.
- By considering symmetry, we can locate the heavy atoms, sometimes uniquely.
- If we know where the heavy atoms are, then we can calculate the scattering factors F_H
- If we know F_H , then we can calculate the phases.

Solving a heavy atom Patterson



Patterson peaks are large circles.

Space group is $P3_1$

Where are the 3 heavy atoms?

- (1) Draw a trigonal unit cell
- (2) Heavy atoms are related by a 3-fold screw, so... draw an equilateral triangle around the origin such that side are Patterson peaks.
- (3) Estimate the coordinates of the triangle vertices.

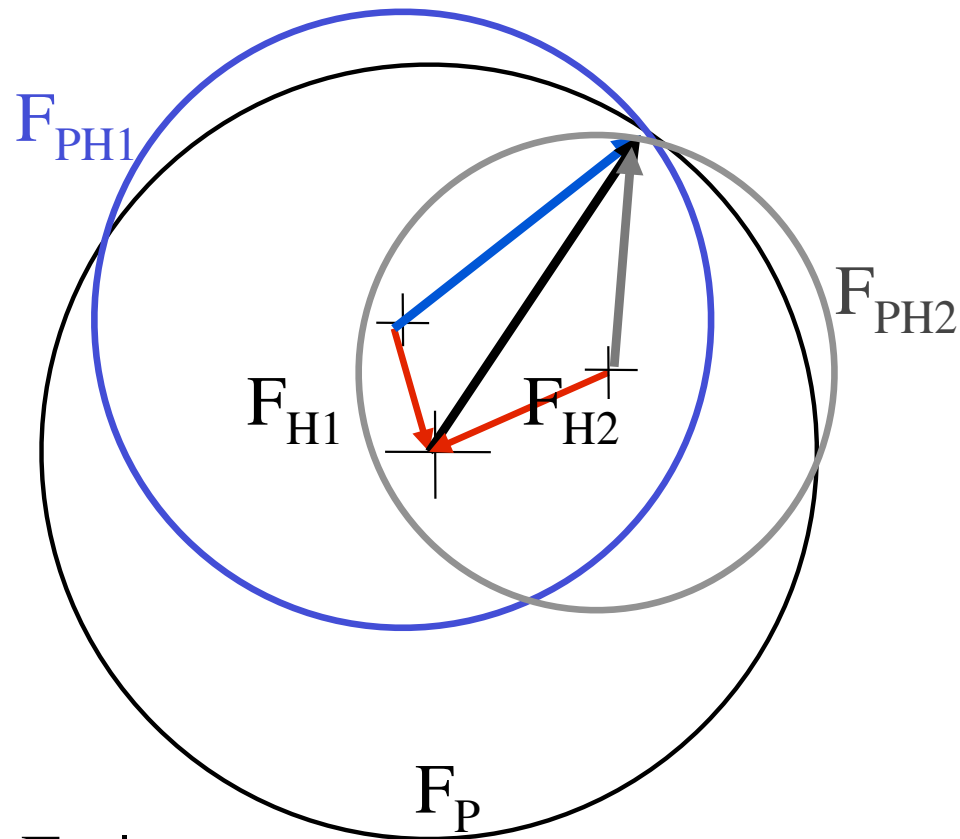
Harker peaks are vectors between symmetry-related heavy atoms.

$$P = v - \underline{M}v$$

Finding and using symmetry operators

- Google International Tables of Crystallography (<http://it.iucr.org/>)
- Volume A
- Choose space group.
- Coordinates.

Two heavy atom derivatives (MIR), unambiguous phases



$$|\mathbf{F}_P + \mathbf{F}_{H1}| = |\mathbf{F}_{PH1}|$$

$$|\mathbf{F}_P + \mathbf{F}_{H2}| = |\mathbf{F}_{PH2}|$$

Or

$$|\mathbf{F}_P| = |\mathbf{F}_{PH1} - \mathbf{F}_{H1}| = |\mathbf{F}_{PH2} - \mathbf{F}_{H2}|$$

Exercise 4: Solve the phase problem for one F using two F_H 's

$$|F_P| = 29.0$$

$$|F_{PH1}| = 26.0$$

$$|F_{PH2}| = 32.0$$

$$F_{H1} = 7.8 \quad \alpha_{H1} = 155^\circ$$

$$F_{H2} = 11.0 \quad \alpha_{H1} = 9^\circ$$

Draw three circles with the three radii (scale doesn't matter)

Offset the PH1 circle from the P circle by $-F_{H1}$

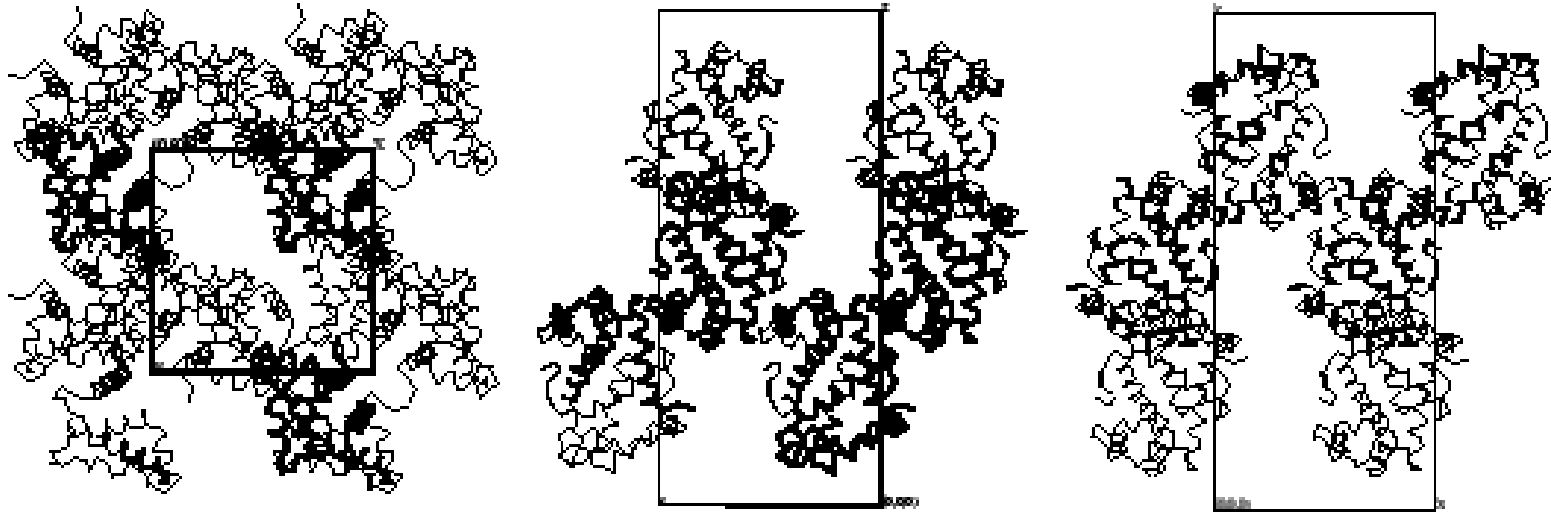
Offset the PH2 circle from the P circle by $-F_{H2}$

Find the intersection of the circles.

Additional topics...

- Crystal packing,
- centric reflections,

Crystal packing



Protein crystal packing interactions are salt-bridges and H-bonds mostly. These are much weaker than the hydrophobic interactions that hold proteins together. This means that (1) *protein crystals are fragile*, and (2) *proteins in crystals are probably not significantly distorted from their native conformations*.

The special usefulness of Centric reflections

- If the crystal has *centrosymmetric symmetry*, all reflections are **centric**, requiring phase = 0° or 180°
- If a non-centric space group has **2-fold**, **4-fold** or **6-fold** rotational symmetry, then the reflections in the ***0-plane*** are **centric**. (Because the projection of the density is centrosymmetric)

For **centric reflections**:

$$|F_{ph}| = |F_p| \pm |F_h|$$

...is *exact**.

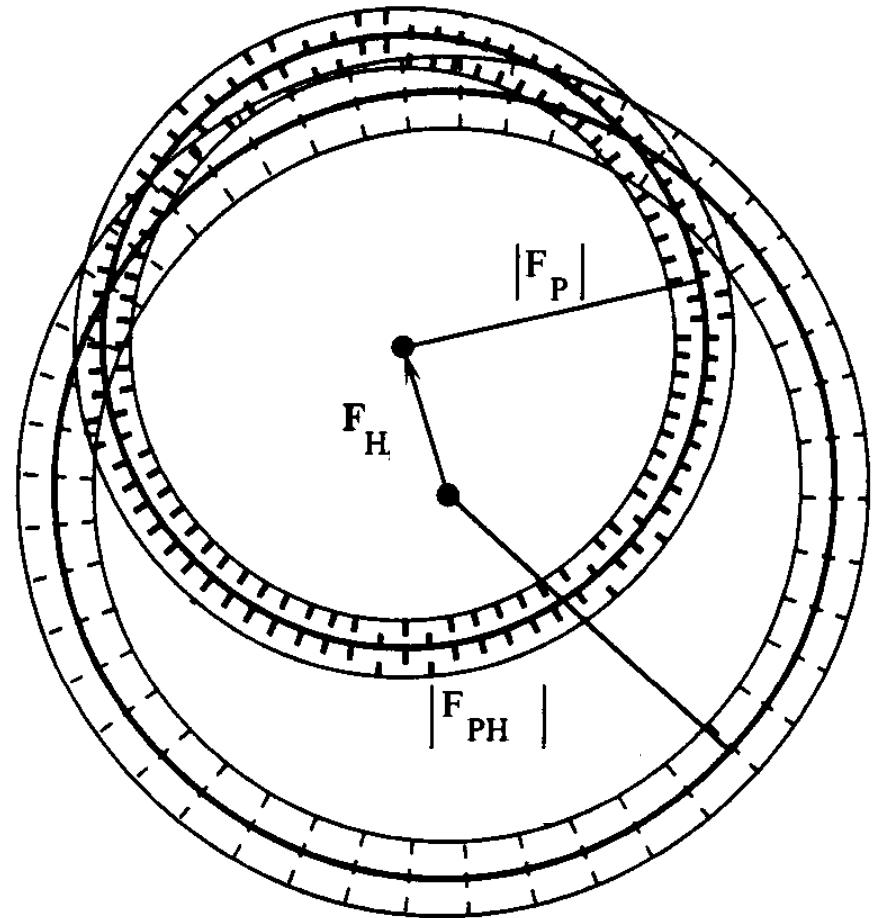
The \pm is + if the phase of F_p and F_h are both 0, or both 180, otherwise -.

*assuming perfect scaling.

Initial phases

Phases are not measured exactly because amplitudes are not measured exactly.

Error bars on F_P and F_{PH} create a distribution of possible phase values α .



width of circle is 1σ deviation, derived from data collection statistics.

Review questions

1. What equation is represented by a Harker diagram?
2. Why is there more anomalous signal at high resolution?
3. What is MAD?
4. What are the three circles in a MAD Harker?
5. What are centric reflections?
6. How do we find the figure of merit (m)?
7. Where is the figure of merit applied in the reverse FT?
8. What gives rise to a phase distribution?

Review questions

9. What does homology mean, in terms of the structure?
10. Are crystals with the same cell dimensions and symmetry and the same contents isomorphous?
11. How to we compare the intramolecular peaks between two Pattersons?
12. What is the rotation function?
13. What is the translation function?
14. What is the R-factor?

Review questions

15. Given a symmetry operator, can you find the equivalent position of a point?
16. What units is Patterson space in?
17. What kind of symmetry does Patterson space always have?
18. What does a peak in Patterson space mean?
19. How is symmetry used to solve a real space position from a Patterson space position?
20. Where are the Harker sections in $P2_12_12_1$?
21. What is the equation for subtracting two data sets (F_P , F_{PH}) to get another data set (F_H)?
22. What does "solving" a Patterson mean?
23. Why can't we measure phases experimentally?
24. What is a Harker diagram? How do you solve it?